

Particles and Fields—Magnetosphere

5705 Ion Shock Waves
S. M. Kasper, University of Maryland, College Park, MD 20742, and C. C. Goodrich, C. W. Moore, and R. F. Winglee, University of Maryland, College Park, MD 20742.
Simulations of a high Mach number shock with parameters typical of the earth's bow shock have been performed using a hybrid particle-in-cell, fluid electron code. The simulation reproduces the observed ion reflection and overdensity in the magnetosheath and magnetopause. These features are shown to be closely associated with ion gyration. (Geophys. Res. Lett., Paper 11272)

5715 Electric Fields
POLAR CAP ELECTRIC FIELD DEPENDENCE ON SOLAR WIND AND MAGNETOSPHERIC PARAMETERS
D. L. Hays, University of Alaska, Fairbanks, Alaska 99701.
A magnetic field model with a linearly superposed uniform electric field is used to study the effect of the magnetospheric field intensity on the relationship between the convection pattern in the polar cap and the latitudinal magnetic field (MLT). It is found that as the magnetic field intensity decreases, the overall electric field intensity decreases, and the dependence of the MLT on the magnetic field intensity becomes more complex. The convection pattern corresponding to "normal" and "reversed" MLT patterns, respectively, is shown. The convection pattern is more complex in the southern hemisphere than in the northern hemisphere. The electric field intensity is more complex in the southern hemisphere than in the northern hemisphere. The electric field intensity is more complex in the southern hemisphere than in the northern hemisphere. (Geophys. Res. Lett., Paper 11272)

5720 Interaction between solar wind and magnetosphere
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5725 Magnetic Field
EXTERNALLY DRIVEN MAGNETIC RECONNECTION VERSUS INTERNAL FIELD RECONNECTION
T. Sato, Institute for Space and Astronautical Sciences, University of Tsukuba, Japan, and A. A. Hasegawa, University of California, San Diego, California 92092.
Theoretical analysis as well as numerical simulations show that while a resistive tearing mode instability is stable at a low frequency, an externally driven reconnection, however weakly driven, brings a new nonlinear state with strong plasma jet which separates the magnetosphere into two regions. This result suggests that the externally induced reconnection is more than likely a candidate for magnetospheric reconnection. (Geophys. Res. Lett., Paper 11274)

5729 Magnetosphere
THE MAGNETOSPHERE AS A SOURCE OF ENERGETIC PARTICLES, MAGNETIC FIELD, AND PLASMA MEASUREMENTS ON NOVEMBER 20, 1977
T. A. Fritz, Space Environment Laboratory, NASA/GSFC, Greenbelt, MD 20899, D. J. Williams, J. Paschmann, C. T. Russell, and V. M. Sotnikov, University of California, San Diego, California 92092.
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A. J. Velli, University of California, San Diego, California 92092.
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PHOTOCHEMISTRY IN PLANETARY ATMOSPHERES

J. S. Levine

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T. E. Graedel

Bell Laboratories, Murray Hill, New Jersey

Theoretical studies of the photochemistry of planetary atmospheres (including those of the ancient and modern earth) are based on physical and chemical principles common to all. Widely varying paths of evolutionary history, atmospheric processes, solar fluxes, and temperature have produced vastly different atmospheres, however. Some of these similarities and differences are described in this paper, which is based in part on invited presentations at the Fall 1980 AGU Meeting.

Introduction

Ozone is a constituent of earth's atmosphere; it is also found in the atmosphere of Mars. Sulfur dioxide is a constituent of the atmosphere of Venus; it is also found in the atmosphere of Io. Methane may have been a constituent of the earth's ancient atmosphere; it is found today in the atmospheres of Titan, Jupiter, and Saturn. Can chemists explain these similarities between systems so different in their physical properties?

Over the last decade and a half, considerable progress has been made in the study of the chemistry of the atmospheres of the planets and their larger satellites (some of the characteristics of which are listed in Table 1). Much of the early research (in the middle and late sixties) was stimulated by NASA's program of exploration of Mars and Venus by the Mariner spacecraft. These early studies were largely concerned with the photochemistry of the carbon dioxide (CO_2) atmospheres of Mars and Venus and of the ability of these predominantly CO_2 atmospheres against photolysis by solar ultraviolet radiation.

In the 1970's, there was a sharp increase in public

concern over issues related to the chemistry of the earth's atmosphere, particularly the impact of anthropogenic activities on the composition of the global atmosphere and on climate. Although for several decades the chemistry of urban air pollution had been a clearly recognized local problem, only recently have the effects of anthropogenic activities on the global atmosphere been recognized as potentially significant. National and international concern has centered on the possible inadvertent depletion of stratospheric ozone (O_3), which protects the surface of earth from potentially harmful solar ultraviolet (200–300 nm) radiation. Atmospheric chemists have studied the possible depletion of stratospheric O_3 by a variety of anthropogenic activities: nitrogen oxides ($\text{NO}_x = \text{NO} + \text{NO}_2$) produced by high altitude supersonic transport, chlorofluorocarbons released from a variety of sources, and enhanced levels of nitrous oxide (N_2O) resulting from increased global use of nitrogen fertilizers.

Atmospheric chemists have also studied the effects of anthropogenic activities on the quality of the air that we breathe, on vegetation, and on the chemistry of natural bodies of water. These studies have included formation and deposition of acid precipitation and the possible effect of enhanced levels of CO_2 resulting from increased burning of fossil fuels (particularly coal) on the earth's climate.

Several anthropogenic activities have been identified as significant sources of various gases to the regional and global troposphere, including CO_2 and sulfur dioxide (SO_2) from fossil fuel burning, NO_x and carbon monoxide (CO) from internal combustion engines, and ammonia (NH_3) resulting from coal conversion and combustion and from the volatilization of nitrogen-containing agricultural fertilizers. Several of these anthropogenic gases initiate chemical reactions that lead to the production of other gases and atmospheric aerosols. Near the ground, CO is involved in a complex series of chemical reactions that leads to the formation of O_3 , which is both a pollutant and an irritant. Anthropogenic NO_x and SO_2 lead to the chemical production of nitric acid (HNO_3) and sulfuric acid (H_2SO_4), respectively, the two dominant acids in rain and snow. Anthropogenic NH_3 , NO_x , and SO_2 lead to the chemical production of atmospheric aerosols (i.e., ammonium nitrate and ammonium sulfate). These aerosols affect visibility and may alter the earth's radiation balance by perturbing the natural ultraviolet, visible, and infrared energy fluxes. During the coming decades, energy generation and industrial and agricultural activities may increase at a very rapid rate, and atmospheric chemists are attempting to predict the effects of such growth on the chemistry of the future atmosphere.

Coincident with these efforts to predict the future, there are looking back in time to study the origin and evolution of the earth's atmosphere. Atmospheric chemists are studying the chemical composition of the early atmosphere and how that composition varied over geological time. They are attempting to deduce the atmospheric conditions on the early earth that were involved in the chemical evolution of the complex molecules (the building blocks of life) and subsequently to biological evolution on our planet.

During the last decade, planetary exploration flourished, with sophisticated interplanetary spacecraft probing the atmospheres of Venus, Mars, Jupiter, Saturn, and the satellites of Jupiter and Saturn. The Mariner, Viking, Pioneer, and Voyager spacecraft transmitted new and exciting information about these planets and provided atmospheric chemists with an unparalleled opportunity to study the chemistry of their atmospheres. Thus, it is not surprising that the last decade has been one of tremendous progress in our understanding of the chemistry of the earth's atmosphere — past, present and future — and of the atmospheres of the other planets. Against this background, the American Geophysical Union organized a day-long symposium on "The Photochemistry of Planetary Atmospheres" during the Fall 1980 meeting.

The symposium was held at the American Geophysical Union headquarters in Washington, D.C., and was attended by over 100 participants. The symposium was organized by J. S. Levine, President of the American Geophysical Union, and T. E. Graedel, President-Elect of the American Geophysical Union. The symposium was held at the American Geophysical Union headquarters in Washington, D.C., and was attended by over 100 participants. The symposium was organized by J. S. Levine, President of the American Geophysical Union, and T. E. Graedel, President-Elect of the American Geophysical Union. The symposium was held at the American Geophysical Union headquarters in Washington, D.C., and was attended by over 100 participants. The symposium was organized by J. S. Levine, President of the American Geophysical Union, and T. E. Graedel, President-Elect of the American Geophysical Union.

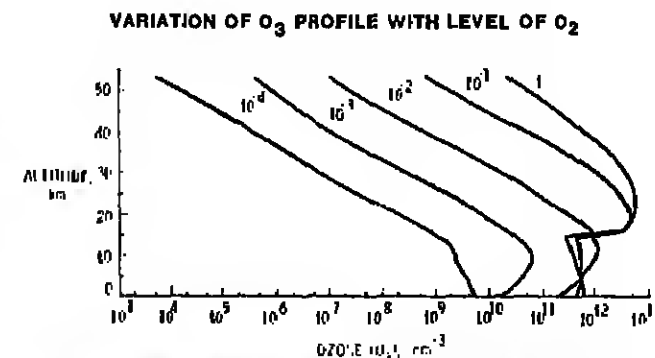


Fig. 1. Theoretical calculations of ozone profiles as a function of molecular oxygen level (in terms of present atmospheric level) in the earth's ancient atmosphere. (J. S. Levine, T. R. Augenstein, R. E. Boughner, M. Natarajan, and L. J. Seaks, in *Comets and the Origin of Life*, edited by C. Ponnamperuma, D. Reidel, Dordrecht, Mass., 1981.)

In San Francisco. In a series of invited papers, 14 atmospheric chemists reviewed the recent progress in their area and identified the problems for future research. This report is a summary of our current knowledge of planetary photochemistry, based in part on the presentations of the symposium.

The Earth's Ancient Atmosphere

Jamaa C. G. Welker (University of Michigan) began the session with a paper entitled "Chemical Evolution of the Atmosphere." From the standpoint of the atmospheric chemist, the early earth is like another planet, with distinct bulk atmospheric composition, solar luminosity, rotation rate, and tectonic properties. Data on the properties of the paleoatmosphere have been preserved, albeit imperfectly, in the sedimentary rock record.

The rock record begins 3.5 billion years (b.y.) ago with highly metamorphosed sediments at Isua in west Greenland. These rocks indicate an early atmosphere containing H_2O and CO_2 and without abundant methane (CH_4). A diverse biota existed by 3.5 b.y. ago, as demonstrated by remains discovered in the Warrawoona Rock Group of western Australia. The continued presence of life sets constraints upon the temperature and composition of the paleoatmosphere.

The bulk composition of the atmosphere — past and present — depends on complex interactions between atmosphere, ocean, crust, and biota. Interpretation of the rock record is providing an increasingly clear picture of how these interactions may have been different on the early earth. Of importance is a gradual increase with time of the area of dry land. At about 2.5 b.y. ago, there was a relatively rapid increase in the area of continental crust.

Oceanic composition was dominated by interaction with the mantle of the early earth. It was a reducing medium, probably saturated with ferrous iron. The continental influence on oceanic composition did not become apparent until the end of the Archean, 2.5 b.y. ago. Theoretical studies of the rates of tidal energy dissipation indicate that the day was about 15 hours long for the first few billion years of earth history.

The geologic record of climatically sensitive sedimentary rocks suggests a climate very different from that of the last few hundred million years of earth history. This information, coupled with increasing solar luminosity, can be interpreted to provide estimates of the partial pressure of CO_2 in the ancient atmosphere (these estimates are a few hundred times the present CO_2 value). The rock record concerning the reduction/oxidation state of the atmosphere and hydrosphere is particularly rich. Oxidized minerals first began to form about 2.3 b.y. ago. By about 1.7 b.y. ago both the atmosphere and hydrosphere contained enough oxygen (O_2) to support aerobic life.

Joel S. Levine (NASA Langley Research Center) reviewed the "Photochemistry of the Paleoa-



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Cover: The 7-m antenna and 0.8-m secondary aperture used at Bell Laboratories, Crawford Hill, New Jersey, for centimeter wave space-echo propagation experiments. These experiments measure radar signal amplitude scintillation caused by turbulence in clouds and depolarization and attenuation caused by solid and liquid hydrometeors. Signal sources were 19 and 28 GHz beams on geosynchronous COMSTAR satellites. The 7-m antenna is also used for millimeter wave radio astronomy. (From D. C. Cox et al., *Observations of cloud-produced amplitude scintillation on 19- and 28-GHz earth-space paths*, *Radio Science*, 16(5), 885-897, 1981.)

	Radius (km)	Solar Dist. (AU)	T (°K)	P (atm)	Main Atmospheric Gases
Earth (Ancient)	6378	1.0	180-300	1.00-1.25	CO_2 , N_2 , CO (CH_4 , PN_2 , H_2)
Earth (Troposphere)	6378	1.0	270-300	0.1-1	N_2 , O_2 (CO_2 , H_2)
Earth (Stratosphere)	6378	1.0	210-270	.001-1	N_2 , O_2 (CO_2 , O_3)
Earth (Mesosphere)	6378	1.0	180-270	10^{-6} - 10^{-3}	N_2 , O_2 (CO_2)
Venus	6200	0.7	600-900	10^{-1} -1	CO_2 (N_2)
Mars	3400	1.5	140-250	10^{-8} - 10^{-6}	CO_2 (N_2)
Jupiter	71400	5.2	110-180	.001-2	H_2 , He
Saturn	60400	9.5	90-160	.001-2	H_2 , He
Uranus	23800	19.2	60-160	.001-5	H_2 , OH_2
Neptune	22300	30.0	60-160	.001-10	H_2
Pluto	3000	29.6	50-80	10^{-4} - 10^{-3}	CH_4
Titan	2580	9.6	~80	~1.5	N_2 (CH_4)
Trifon	2500	30.0	~80	~ 10^{-4}	H_2 , N_2 , Ar , CH_4

sphere. As was previously noted, the chemical composition and evolution of the paleoatmosphere were controlled by a strong coupling between the atmosphere, the oceans, the solid earth, and, eventually, the biosphere. The composition was also modified by various atmospheric processes, including photochemical reactions (initiated by the action of solar ultraviolet radiation, which was considerably more intense in the O_2 -deficient paleoatmosphere), chemical reactions, lightning, rainout, and the exospheric escape of light atmospheric gases. Photochemical studies of the paleoatmosphere can be divided into three main areas of research: (1) the photochemistry and stability of the early anoxic atmosphere, (2) the chemical transition to an oxidizing atmosphere, and (3) the origin and evolution of atmospheric O_2 , with the accompanying shielding of the earth's surface from lethal solar ultraviolet radiation.

Since the early laboratory experiments on chemical evolution, in which complex organic molecules (the precursors of living systems) were synthesized in mixtures of NH_3 and CH_4 exposed to ultraviolet radiation or laboratory electric discharges, it became fashionable to believe that the prebiological primitive atmosphere contained large amounts of NH_3 and CH_4 . However, photochemical calculations indicate that such an early prebiological atmosphere would have been highly unstable against photolysis by solar ultraviolet radiation and, hence, would have been very short lived, on a geological time scale. If it ever existed at all, in addition, there is no geological or geochemical evidence in the rock record to support such a highly reducing early atmosphere. A more mildly reducing atmosphere of N_2 , CO_2 , CO , and H_2 (resulting from volcanic outgassing) is now favored by photochemical, geological, and geochemical considerations. Complex organic molecules have been synthesized in such laboratory mixtures. Recent photochemical calculations also indicate that large amounts of nitriles (formed by lightning) and formaldehyde (formed by atmospheric reactions) could have been transported to the early ocean by precipitation in the O_2 -deficient paleoatmosphere.

The transition from a reducing paleoatmosphere to an oxidizing atmosphere resulted from the build up of atmospheric O_2 . Photochemical calculations indicate that the photolysis of H_2O , with the accompanying exospheric escape of H , was probably not a significant source of O_2 over geological time. The inefficient exospheric escape of H (comparable to today's value) and the volcanic outgassing of H_2 led to a efficient reformation of H_2O , at the expense of O_2 build up. In the prebiological paleoatmosphere, O_2 was not evenly mixed with altitude, but had a concentration of about 10^{-12} present atmospheric level (P.A.L.) or less at the surface, and a maximum concentration of about 10^{-8} P.A.L. at about 40 km. It appears that photosynthetic activity was the major source of atmospheric O_2 , although there is considerable uncertainty as to the exact chronology for the build up of atmospheric O_2 over geological time.

Recently, a great deal of research has centered on the origin and evolution of O_3 , which was strongly coupled to the build up of O_2 (see Figure 1). The evolution of O_3 and the variation of solar ultraviolet radiation reaching the surface of the earth over geological time (which is controlled by O_2) may have had very important implications for the origin and evolution of life on our planet. Studies on the evolution of O_3 have been based on detailed one-dimensional tropospheric/atmospheric photochemical models that include the chemistry of the oxygen, nitrogen, hydrogen, carbon, and chlorine gases. These studies have shed new light on the photochemistry of the paleoatmosphere (for O_2 levels of 10^{-4} P.A.L. to the present). In the O_2 -deficient paleoatmosphere, the enhanced level of solar ultraviolet radiation rendered photolytic destruction the primary loss mechanism for N_2O , N_2 , and O being the products. (In the present atmosphere, N_2O is largely oxidized to NO_x , thus becoming involved in the stratospheric ozone cycle.) The enhanced levels of solar ultraviolet radiation in the O_2 -deficient paleoatmosphere also resulted in the efficient production of OH via the photolysis of H_2O resulting in surface and atmospheric levels of OH several orders of magnitude greater than in the present atmosphere. The sensitivity of paleoatmospheric O_3 to varying values for solar luminosity, atmospheric temperature, vertical eddy transport, and trace atmospheric gases have been assessed in these studies.

The Earth's Present Atmosphere

T. E. Graedel (Bell Laboratories) discussed the "Photochemistry of the Regional Troposphere." The chemistry of the regional troposphere (a regime defined as encompassing distance scales of the order of 10–1000 km) is intertwined with, but not dominated by, meteorological motions and local emission sources. The time scales of the air motions prescribe the chemical lifetimes of interest, ~1 hour to 5 days, and thus define the species whose chemistry must be studied. These include ammonia, the oxides of nitrogen, the sulfur-containing compounds hydrogen sulfide (H_2S) and sulfur dioxide (SO_2), and many alkenes, terpenes, aromatics, and aldehydes.

The following are among the regional tropospheric problems of current interest:

1. Downwind Effect. The concentration of a variety of photochemical products are known to be higher downwind than in the vicinity of the precursor,

Forum

Trend Toward Multiple Authors in Research Publications: Failure of the Universities to Support Research

In a previous letter to this column (Fraser-Smith, 1979) one of us drew attention to the marked decline since 1950 in the percentage of single-author papers in the *Journal of Geophysical Research*, *Space Physics* (JGR 1) and the commensurate increase in the percentage of articles by three or more authors. The decline in single authors is certainly not confined to JGR 1, as is shown by more recent work (according to the Institute for Scientific Information, which indexes 2800 journals, the average number of authors per paper rose from 1.67 to 2.58 between 1960 and 1980) (Brod, 1981), so it seems clear that there is a widespread change taking place in the way scientists report the results of their research. It is perhaps important for us to point out that this is not an academic change; it is taking place right now, and most readers of this column are likely to be affected by it.

There is undoubtedly an element of fashion involved in the decline of single authors, and it may well be that what we are all experiencing individually as scientists is a subtle process of rhinocerotization, as described in the play by Ionesco (1960). However, it would be unusual for a fashion to persist for 30 years or more unless other more substantive factors were involved. The question is, what are these factors?

It would be easy to blame the federal government for the change that is occurring in our reporting habits (see Price, 1981) since funding of research by federal agencies has become significantly less in the 1950's and, as we all know, it has grown remarkably since. However, an extension of the earlier work on authors, as suggested to us by C. T. Russell, indicates that the changes in the numbers of authors per article are not linked directly to the growth in federal funding. In fact, we will argue that a more important reason for the decline in single authors is a lack of support by our universities.

Following the suggestion by Russell, we reanalyzed our multiple-author data for JGR 1 according to the acknowledged sources of support for the work. Needless to say, our new data are less quantitative than before, but the trends, as illustrated in the figure, are probably accurate. Shown in the figure are the variations since 1950 in the percentage of papers acknowledging NASA, NSF, or ONR support (these are the three most frequently acknowledged agencies in JGR 1), or no support. Note that it is the percentages of papers in each category that are plotted (one, two, three or more authors), and thus the trends in the number of authors per article previously discussed should not be evident. Clearly, the percentage of papers with no acknowledgment of support has decreased substantially in every author category since 1950, with the most marked decrease occurring in the papers by either one or two authors. Sady, the percentage of papers acknowledging support from ONR, the first of the U.S. government research funding agencies, has also declined. However, NASA and NSF support has increased substantially and has more than made up for the decline in ONR support. The most interesting feature of the increase since 1980 in the percentage of papers acknowledging support from NASA and NSF is that it takes place in all author categories. In other words, the advent of large-scale federal funding cannot be held directly responsible for the precipitous decline in single-authored papers since 1950.

A possible clue to another cause of the decline is obtained by comparing the acknowledgments in single-authored papers for 1980 and 1981. It was rather common in 1980 for single authors affiliated with U.S. universities to omit acknowledgment of a source of funds for their research (29% of the relevant authors fell into this category; the percentage is even greater in earlier years), whereas in 1980, essentially all such authors acknowledge a source of funds outside their university (only 3% fail to acknowledge support). This result is open to a variety of interpretations, but it appears that an entire class of researchers may have disappeared in the interval 1980–1981: the university researcher, usually a tenured professor, supported solely by university funds. It is this case, a prime source of diversity and originality in research has undoubtedly been lost. One has only to read the recent comments by Willenbrock (1981) on the decline in U.S. technological leadership to realize that this loss has implications beyond the mere authorship of scientific papers.

We will not dwell on the advantages of U.S. government research contracts and grants to the universities or on the pressures that are exerted on university faculty and staff to bring in federal funds for research. Instead, we wish to point out that there is an alternative but neglected way to support research at universities. Once again the clue is provided by the acknowledgment sections of the papers we have analyzed. On a very few occasions, support from research funds administered by a university is acknowledged.

amissions. This effect, an example of which is shown in Figure 3, is qualitatively understandable as a consequence of the interplay between the chemical reaction times of the emissions and the wind velocity. Recent model studies show promise of being able to reproduce this effect quantitatively as well. Detailed measurements are needed to guide further model development.

2. Fate of Terpenes. Terpenes are emitted from vegetation in large, though uncertain, amounts. It has

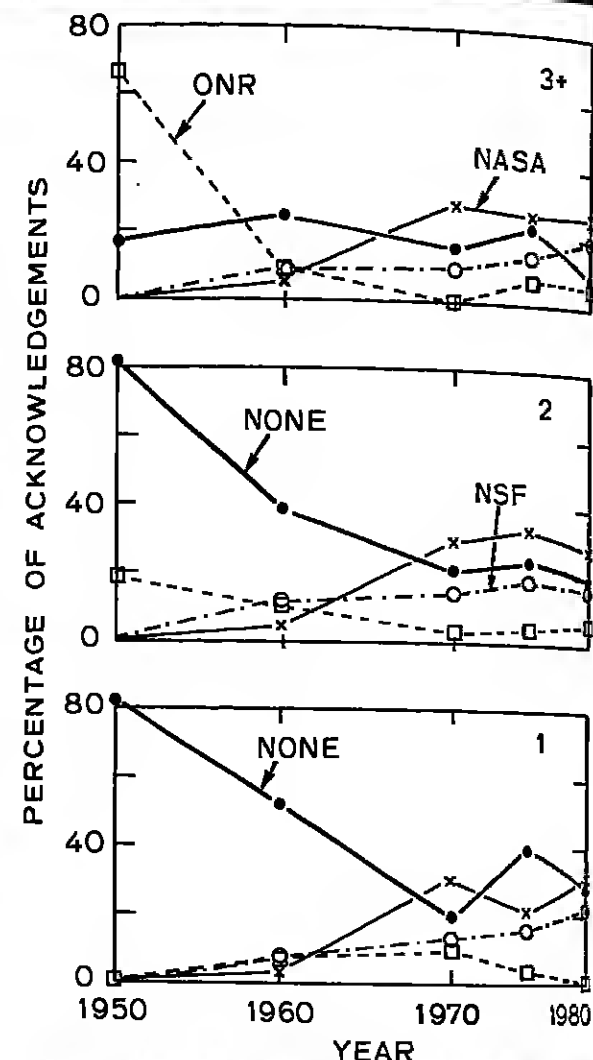


Fig. 1. Variation since 1950 in the acknowledgments of NASA, NSF, and ONR financial support in papers with one (bottom panel), two (middle), and three or more (3+) (top panel) authors published in the *Journal of Geophysical Research*. Also shown is the variation in the proportion of papers with no acknowledgment of financial support. The percentages that are plotted apply to each author category.

edged [e.g., Schwind et al., 1980; Rankin and Kurtz, 1979]. We might ask why these acknowledgments are so slow. It appears that this ready availability of federal funds for research in the past has blinded university research administrators to the advantages of a general research fund, established and increased by gifts and bequests, whose income is used solely to fund internal (or even external) research proposals. Some of the strings normally attached to federal funds, long bemoaned by these administrators, can be avoided, and a wider range of research can be undertaken, in keeping with the particular objectives of the universities. Most important, the disturbing possibility that the federal funds for particular programs of research can be eliminated by the efforts of small groups of congressmen, or even a single U.S. senator, as appears to be happening now in the case of the NASA-funded Search for Extraterrestrial Intelligence (SETI) program, can be more easily avoided.

It might be argued that U.S. universities cannot afford to support research out of their own funds. We disagree. Large sums are raised each year from alumni, companies, and other sources, particularly at what are termed the "major research universities" (i.e., universities that have been particularly successful at soliciting federal research funds), and even a small diversion of these funds each year into a university research fund (specifically earmarked for research) could soon produce significant income.

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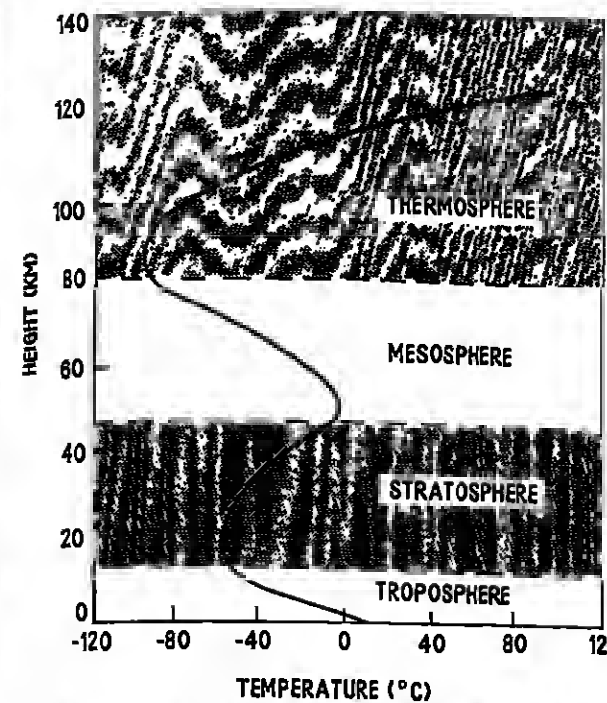


Fig. 2. The four principal layers of the earth's present atmosphere. The boundaries between the layers fluctuate somewhat with time and with geographical location.

meteorological motions, with apparently dramatic chemical consequences in some cases. Efforts to model these processes by combining detailed descriptions of organic and inorganic photochemistry and of boundary layer meteorology are in their infancy.

4. Aerosol Nucleation and Growth. Chemical evidence from airborne particulate matter implies that a rich chemistry occurs on particles as a result of their nucleation and growth from reactive gas molecules. Our understanding of these processes is poor and appears limited at present by theoretical uncertainties more than by lack of data, although the latter are surely scanty.

5. Acid Rain. It seems clear that oxides of sulfur and nitrogen emitted in the gas phase are responsible for the increasing acidity of precipitation in northern Europe, northeast United States, and other areas. The rates at which the gas-to-drop transition occurs, and, in fact, the mechanisms of that occurrence, are poorly known. A marriage of atmospheric chemistry, aerosol physics, and cloud physics may be required to reduce the problem to quantitative understanding.

The regional troposphere is probably the most chemically diverse of any of the planetary atmospheric regimes. Perhaps more than any other regime, it must blend emissions, meteorology, and chemistry to analyze properly the processes that occur. Its study is an example of the interdisciplinary requirements of modern atmospheric and planetary science.

William L. Chambliss (Georgia Institute of Technology) reviewed the "Photochemistry of the Global Troposphere." Key elements of the tropospheric photochemical system are (1) the production of the free radical OH in the presence of solar radiation and (2) the emission of reduced gases from the earth's surface. Tropospheric OH triggers the oxidation of many of the reduced gases generated at the earth's surface and ultimately causes their transformation into chemical forms that are readily removed from the atmosphere by rainout and other heterogeneous processes. Thus, tropospheric photochemistry supplies the atmospheric link in the biogeochemical cycling of elements such as C, N, and S. In addition, this chemical system can perturb life systems by influencing important environmental parameters such as surface temperature (via the atmospheric "Greenhouse Effect"), the acidity of rainfall (via the production of and incorporation in preprecipitation of H_2SO_4 and HNO_3), and the concentrations of potentially toxic species.

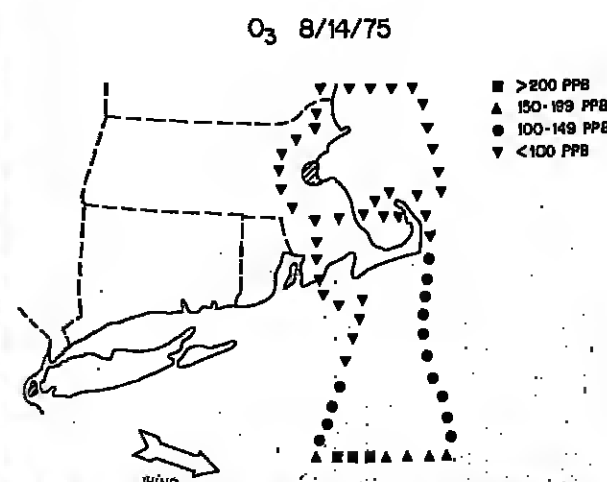


Fig. 3. Ozone concentrations at 330–490 m altitude off the coast of the northeastern United States on the afternoon of August 14, 1975. The highest concentrations were seen about 280 km east of the New York City metropolitan complex, a region of high precursor emission fluxes. Trajectory analysis demonstrated that the high ozone air mass passes over the metropolitan area during the morning of the day on which measurements were made. (After G. W. Stille, C. K. Fitzhugh, D. F. Zeller, and R. B. Evans, EPA-600/3-77-001a, Environ. Prot. Agency, Research Triangle Park, N.C., 1977.)

While many uncertainties remain, over the last decade great strides have been made in our understanding of the atmospheric cycles of the carbon compounds CH_4 and CO , nitrogen oxides, ozone, and halogens. In each of these cycles, it has been found that OH plays a major role, and thus understanding the processes that control the OH abundance is of major importance. Because of its reactivity, the concentration of OH at any location is determined by a balance between photochemical production and destruction. OH production occurs as a result of the photolysis of O_3 to produce a metastable oxygen atom ($O(^1D)$) followed by the reaction of $O(^1D)$ with H_2O . Reactions of OH with CO and CH_4 are generally the major loss processes for OH in the remote troposphere. O_3 and NO can lead to a further enhancement in OH levels by regenerating OH from HO_2 , the HO_2 having been produced by the CO and CH_4 reactions with OH. Thus, an understanding of OH concentrations in the atmosphere requires a corresponding understanding of the atmospheric distributions of O_3 , H_2O , CO , CH_4 , the nitrogen oxides, and perhaps the array of atmospheric hydrocarbons.

The recent National Science Foundation-sponsored Global Atmospheric Measurements Experiment (GAME-TAG) included simultaneous measurements of OH and many of the parameters that influence OH concentrations: CO , CH_4 , O_3 , H_2O , ultraviolet flux, etc. Thus this experiment, by allowing a direct comparison of measured and model-calculated OH, has made possible the first quantitative test of the photochemical theory of tropospheric OH. Results for the tropical and subtropical marine boundary layer are quite encouraging. It is suggested, however, that a future project consisting of a more comprehensive set of trace gas measurements would afford a more rigorous test of OH chemistry and of other fast photochemical processes. It is also noted that a vigorous observational program to elucidate the detailed global distributions of key trace gases and their sources and sinks is still a major need in this field, as is the continued development of sophisticated photochemical models to analyze and simulate these data.

Berry J. Hubert (Colorado College) considered the "Aerosol Chemistry of the Troposphere." Because so many gases that cycle through the troposphere pass through a condensed phase at some point, we cannot fully understand their geochemical cycles without understanding their aerosol phases. Current areas of aerosol research include (1) measurements of particle sizes, compositions, and spatial distributions; (2) studies of the effect of aerosols on visibility, climate, and precipitation chemistry; and (3) investigations into the role aerosols play in trace gas chemistry.

In many cases the aerosol phase is a sink for trace gases. This gas-to-particle conversion includes such processes as the nucleation of new particles (after the reaction of SO_2 with OH, for instance), the condensation of gases onto existing particles (as in the adsorption of HNO_3 onto aerosols), and the heterogeneous reaction of adsorbed gases on particles that serve as catalysts. Aqueous phase chemistry can occur in the liquid layer surrounding solid particles in mist, fog, and clouds.

Aerosols are not only sinks for gases, they can serve as a source when volatile gases are formed on or in the aerosol (as in the release of HCl from sea spray) or when cloud droplets evaporate and release their dissolved gases.

We are only just beginning to appreciate the complex role of aerosols in the chemistry of the troposphere. Although we qualitatively understand the processes by which aerosols interact with atmospheric gases, our quantitative understanding is quite poor. Some models for aerosol growth do give fair agreement with chemically simple systems, but current models are generally unable to predict the relative rates of nucleation, condensation, and coagulation of the complex aerosols in the real atmosphere. We need better thermodynamic data for the impure condensed phases that form, as well as much additional theoretical and experimental work on the growth processes. Until we can predict gas-to-particle and particle-to-gas conversion rates, we will be unable to include meaningful source and loss terms in atmospheric trace-gas models. The situation is only slightly better for the measurement of aerosol composition and concentration. Although some non-volatile, noncondensable compounds can be collected and analyzed directly, many aerosol measurements are hampered by positive or negative artifacts from the sampling process. Aerosol chemistry desperately needs techniques that can identify specific chemical species in situ aerosols.

Ralph J. Ciccone (National Center for Atmospheric Research) summarized the "Photochemistry of the Stratosphere" (the region of the atmosphere between 10 and 60 km above the surface). The stratospheric chemistry is controlled by the fluxes of gases from the troposphere and the mesosphere and by their interaction with solar ultraviolet and visible radiation. A major concern of scientists is the photochemistry of stratospheric O_3 , and the possible inadvertent depletion of O_3 owing to various anthropogenic activities (e.g., high flying supersonic transport, man-made chlorofluoromethanes (CFMs), and nitrogen fertilizers used in agriculture). About 90% of the total atmospheric O_3 is found in the stratosphere. As was already pointed out, stratospheric O_3 protects the surface of our planet from solar ultraviolet radiation (200–300 nm). The production of stratospheric O_3 is initiated by

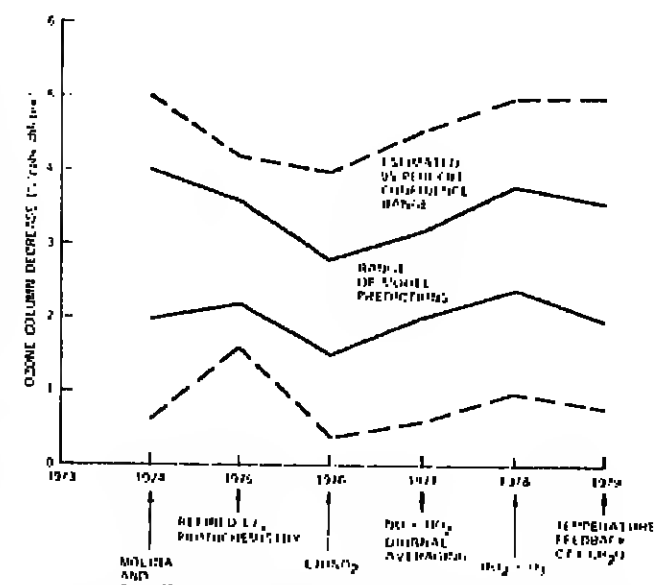


Fig. 4. Theoretical predictions of the eventual decrease in stratospheric ozone as a function of the time at which the prediction was made (courtesy of R. P. Turco).

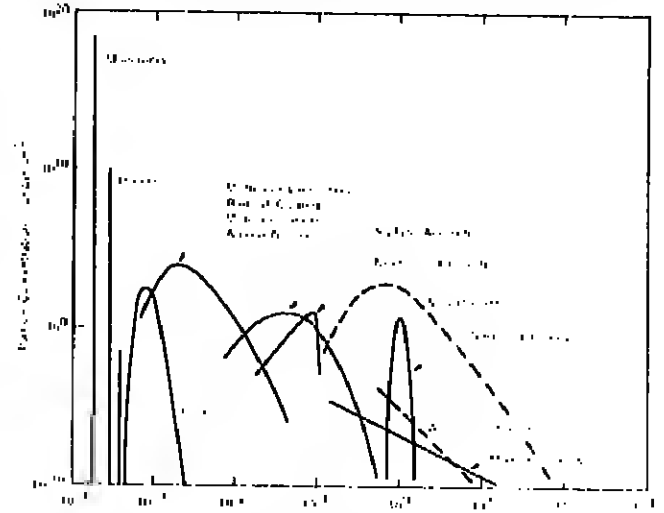


Fig. 5. The spectrum of particles in the earth's upper atmosphere. Shown are the approximate size distributions of particles of different origins. The total number concentration of each type of particle is roughly indicated by the peak value indicated on the vertical scale (courtesy of R. P. Turco).

the photolysis of molecular oxygen (O_2) by solar ultraviolet radiation ($\lambda < 242$ nm), resulting in the formation of two oxygen atoms (O). The oxygen atoms recombine with O_2 to form O_3 . Stratospheric O_3 is destroyed via photolysis and by reaction with atomic oxygen. In addition, stratospheric O_3 is destroyed by reactions with species present in trace amounts, thousands of times less abundant than O_3 itself. These species enter into catalytic cycles in which one molecule can destroy many O_3 molecules before being removed itself. These catalytic cycles involve the oxides of nitrogen (NO_x), hydrogen (HO_x) and chlorine (ClO_x). The major source of stratospheric NO_x is the reaction of nitrous oxide (N_2O) with excited atomic oxygen. Nitrous oxide is produced by the action of microorganisms at the surface of the earth. The major source of HO_x in the stratosphere is the reaction of excited atomic oxygen with water vapor (H_2O). The major source of stratospheric ClO_x is the photolysis of man-made chlorofluoromethanes (CFMs) (e.g., $CFCl_3$ and CF_2Cl_2). Current model results suggest that the CFMs produce the largest ozone depletion. As updated information becomes available, model results fluctuate, however, as is shown in Figure 4. Inadequate computational power and the sparseness of stratospheric data will combine to make improved assessment difficult over at least the next few years.

"Particulates in the Middle Atmosphere" was the topic discussed by Richard P. Turco (R&D Associates). The middle atmosphere—the stratosphere and mesosphere between 15 and 80 km—is now known to contain a rich variety of particulate matter (Figure 6). Neutral and charged clusters of molecules, with radii ranging from 2 to 10 Å, are the most diminutive of atmospheric particles. Clustered positive and negative ions may combine to form stable multi-ion complexes with sizes between 10 and 50 Å. Meteoric smoke, which is produced by the ablation of interplanetary debris entering the upper atmosphere at high velocity, also occupies the small size range from about 5 to 100 Å. Micrometeorites that survive atmospheric entry dominate the size spectrum from about 1 to 10 µm. Sulfate aerosols of photochemical origin are present in a narrow layer between 15 and 25 km altitude. These sulfate particles can grow to sizes of 0.1 to 1 µm. Volcanoes occasionally inject large quantities of ash and sulfur-bearing gases into the middle atmosphere. Such events can significantly increase the aerosol burden for a year or more.

Men can affect the concentrations of aerosols in the middle atmosphere both by direct injection of particulates and by emission of gases that can condense into particles. Aluminum oxide dust from rocket engines has been found in large quantities at 20 km. Some industrial gases such as arsenic trioxide and carbon disulfide can reach the stratosphere, then

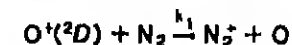
decompose, and subsequently form new sulfate aerosols.

A number of possible mechanistic routes have been suggested for the particles in the middle atmosphere. They may (1) serve as reaction sites for gases, (2) act as a sink for gases, (3) influence the global radiation balance, (4) nucleate nucleation and noctilucent clouds, and (5) provide a link between aerosol emission variations and tropospheric weather. Of these possibilities, only the second and third are well established at present, although the others are under active experimental and theoretical investigation.

Since the discovery of the permanent, ubiquitous stratospheric sulfate aerosol layer by Jung and coworkers two decades ago, these particles have been a subject of intense study using both in situ and remote sensing techniques. On the basis of such observations, we can say that the aerosols are probably composed of a 75% sulfuric acid aqueous solution, with an admixture of a variety of materials ranging from meteoric debris to nitrooxy compounds derived from stratospheric NO_x . The mode of nucleation of the aerosols is uncertain and may involve tropospheric (Aitken), meteoric, ionic, and sulfur radical nuclei. The growth of the aerosols is largely controlled by the oxidation of sulfur-bearing gases which are transported into the stratosphere from the troposphere. The primary gaseous aerosol precursors are sulfur dioxide and carbonyl sulfide, both of which may have significant anthropogenic sources.

The effects of stratospheric aerosols on the global radiation balance are most noticeable following major volcanic eruptions, when the aerosol layer may be enhanced to the point that substantial temperature variations occur at the earth's surface. The May 18, 1980, eruption of Mt. St. Helens created darkened skies and brilliant sunsets, dramatically demonstrating the optical perturbations that can be caused by volcanic aerosols. Recently, on increasing trend in the optical density of the background (non-volcanic) aerosol layer has been identified; one possible explanation is man's increasing usage of fossil fuel, with the attendant release of sulfur compounds to the atmosphere.

In a talk entitled "Photochemistry of the Mesosphere and Thermosphere," Douglas G. Torr (Utah State University) pointed out the influence of new data, largely acquired or stimulated by satellite experiments, on the understanding of thermospheric chemistry. At the end of 1979, this chemistry appeared to have crystallized into a clearly understood form, mainly as a result of analyses conducted with data taken by the Atmospheric Explorer (AE) C, D, and E satellites. To maintain consistency with laboratory measurements of the processes that destroy the N_2^+ ionization, it was found that the rate coefficient for the charge exchange reaction



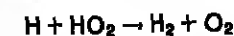
must be less than $10^{-10} \text{ cm}^3 \text{ s}^{-1}$, that is, an order of magnitude smaller than earlier laboratory measurements conducted by using nonthermal O^+ ions. However, during the course of the last year new laboratory measurements established k to be $\sim 8 \times 10^{-10} \text{ cm}^3 \text{ s}^{-1}$ (i.e., an order of magnitude larger than the aeronomically deduced value). Inclusion of this new value for k in the ion chemistry introduced several problems, namely excess production of N_2^+ ions and a shortfall in production of $\text{O}^+(^4S)$ ions, which caused a deficiency in the concentrations of both NO^+ and O_2^+ ions. These appear now to have been simultaneously resolved with the inclusion in theoretical models of rapid charge exchange between vibrationally excited N_2^+ and oxygen atoms. The previously anomalous seasonal behavior of the F_2 layer on a global scale appears also to be explained by this new approach.

Torr reported that a major step forward in the laboratory measurement of rate coefficients was that of E. C. Zipf (University of Pittsburgh), who used laser induced fluorescence techniques to study the behavior of N_2^+ ions in specifically identified vibrational and rotational quantum states. The specific dissociative recombination coefficients for the $v=0$, 1, and 2 vibrational levels of the $\text{N}_2^+(X^2\Sigma_g^+)$ state were found to be nearly equal in magnitude, thereby placing significant constraints on allowable thermospheric models. In the case of O_2^+ Zipf used indirect techniques to study the production of $\text{O}^+(^4S)$ via O_2^+ recombination and found very little production in the $v=0$ level, implying that high $\text{O}^+(^4S)$ yields are associated with recombination in high vibrational levels. The $\text{O}^+(^4S)$ yield inferred from the AE data is large, but vibrationally excited O_2^+ is believed to be strongly quenched by atomic oxygen in the thermosphere. To explain this inconsistency, a dependence of the $\text{O}^+(^4S)$ yield on electron temperature has been tentatively suggested.

In the area of neutral thermospheric chemistry, recent developments include the finding that the forbidden predissociation of the numerous N_2 and O_2 valence and Rydberg states of N_2 in the 11–24 eV range via radiation entrapment in an optically thick atmosphere is the dominant mechanism for N_2 atom production, the finding that destruction of $\text{N}(^2D)$ constitutes a major source of OI 8446 Å dayglow, and the discovery of an oxygen geocorona of ~ 3000 to 4000°K .

Mesospheric chemistry has not received as much recent emphasis as has thermospheric chemistry, mainly because the necessary experimental effort has been planned for the 1980's. Some recent results

have provided new insights, however. A reevaluation of the O_2 dissociation rate has yielded ratios of 1 and 0.6 for the old to new rates at the altitudes 80 and 50 km, respectively. Similar results for the dissociation rate for water vapor indicate that the uncertainties are such that the actual rates may be as low as 0.45 or as high as 1.55 times the currently accepted values. In an analysis of solar proton events, it has been recently pointed out that above ~ 75 km the rate of dissociation of water vapor produced by recombination of hydrated ions may be large enough to deplete H_2O concentrations significantly, since the odd hydrogen at these altitudes is effectively lost in conversion of H_2 by the reaction



The net result would be an initial depletion followed by a storm time increase in ozone.

Terrestrial Planets: Venus and Mars

Planetary atmospheric chemistry at the symposium was introduced by Ronald G. Prinn (Massachusetts Institute of Technology), who discussed the "Chemistry of the Atmospheres of Venus." The atmosphere is dominated by CO_2 , N_2 at $\sim 3.4\%$ being the next most abundant constituent. HCl and HF were detected by ground-based spectroscopy in 1980; their presence is consistent with what one would derive by heating earth rocks to the 750°K Venusian surface temperature. CO is also present, primarily as a result of CO_2 photolysis; other suggested sources are lightning and thermochemistry near the surface. Water vapor is present at concentrations of a few parts per million above the clouds and a few hundred parts per million below the clouds.

The recent Venere and Pioneer Venus probes confirmed earlier suggestions of a rich sulfur chemistry in the Venusian atmosphere by detecting SO_2 and, more tentatively, H_2S , S_2 , and S_3 below the clouds. Collaborative evidence for concentrated sulfuric acid as a major component of the clouds at Venus was also provided. The ultimate source of the sulfur is undoubtedly outgassing of the crust, perhaps partially by volcanism. This outgassing is expected to be in the form of H_2S and COS . The latter gas has not yet been observed but would be expected at altitudes below 20 km (where measurements are presently lacking) due to equilibration of the observed CO , S_2 , S_3 , and SO_2 .

It appears that photochemical oxidation of SO_2 and, to a lesser extent, H_2S is the major source of the sulfuric acid in the clouds and the major sink for the O_2 produced from CO_2 photodissociation at high altitudes. The major sink for CO appears to be oxidation to CO_2 by reaction with SO_2 , SO_3 , and SO near the surface. The Venere spectrophotometer indicated that the water vapor mixing ratio decreases as one approaches the surface. This property may be linked to photodissociation of S_2 and S_3 by near ultraviolet end visible light below the clouds. Collaborative laboratory studies are required.

The chlorine chemistry of the Venus atmosphere, although studied for some years, is more poorly understood than that of the earth. The 1 ppm of HCl that is present will photolyze to produce both odd hydrogen and odd chlorine radicals. These latter species are expected to play some role in the oxidation of SO_2 at cloud level, but the exact mechanisms are not yet known. The much lower O_2 concentrations on Venus will render Cl much more abundant than ClO, in contrast to the situation in the earth's atmosphere.

The clouds of Venus possess a complex structure as a function of altitude (Figure 8) and appear to be composed of several different chemicals as well. These points were discussed by Owen B. Toon (NASA Ames Research Center) in his talk "Chemistry of the Clouds of Venus." The upper clouds are largely sulfuric acid, but the particle size distribution is bimodal with peaks at ~ 0.1 and $1.0 \mu\text{m}$, a distribution that pure sulfuric acid particles apparently cannot reproduce. This implies the presence of another constituent. Sulfur particles violate several observational constraints and cannot provide the opacity needed to explain the ultraviolet markings on Venus. A candidate compound is Cl_2 .

The lower clouds are characterized by a trimodal

VERTICAL STRUCTURE OF VENUS CLOUD SYSTEM

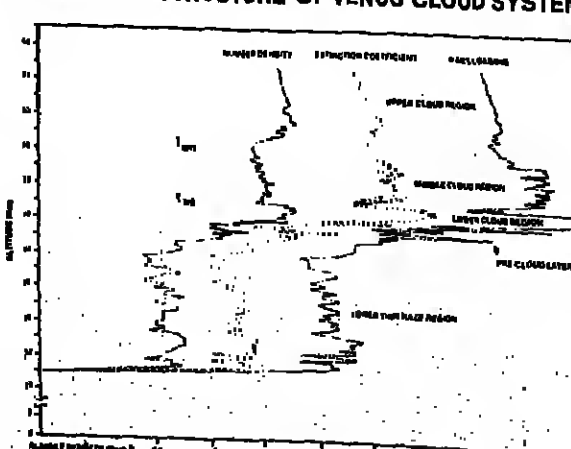


Fig. 8. Particle number densities, extinction coefficients, and mass loadings of the clouds of Venus (courtesy of R. Kieffer).

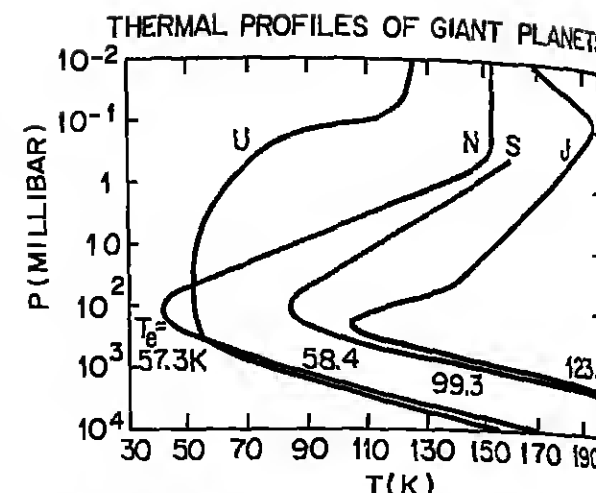


Fig. 7. The thermal structures of the Jovian planets. (D. Gautier and R. Courtillot, *Icarus*, 38, 28, 1978; reproduced by permission of Academic Press.)

particle size spectrum with peaks at ~ 0.01 , 0.1 , and $0.6 \mu\text{m}$. Again, sulfuric acid seems to be controlling species in such a trimodal regime. The clouds appear to be partly sulfuric acid, but most of their mass is contained in an unidentified solid, transparent compound composed of at least Cl, O, and S.

Several difficulties are encountered when trying to model the cloud chemistry. Present models are reasonably successful at reproducing the observed upper level sulfuric acid clouds, but are unsuccessful at producing the observed vertical profiles of O_3 , CO , and SO_2 , the abundance of Cl_2 required to explain the observed ultraviolet absorption, or any solid chlorine compounds in the lower clouds. Theoretical and laboratory studies are required to improve the models, but major questions remain for future spacecraft missions to resolve.

Next to the earth's present atmosphere, that of Mars is probably the best understood. In a talk entitled "Photochemistry of the Martian Atmosphere," T. Y. Kong (Ball Laboratories) pointed out some of the distinctions between them. The atmosphere of the earth is controlled by four processes: photochemical, physical, biological, and anthropogenic. On Mars, only the first two of these appear to exist. As a result, Mars has not developed an oxygen-rich atmosphere, but maintains a thin atmosphere (surface pressure 8 mbar) dominated (98%) by CO_2 . About 4% of the atmosphere is N_2 and Ar.

The low opacity of the Martian atmosphere (except during dust storms) promotes photochemical processes, with such species as O_3 , CO , and NO being observed as a result. Much of the driving force for this chemistry comes from the photodissociation of CO_2 and H_2O . Model calculations have been reasonably successful at treating such features as the wide variation (up to a factor of ~ 30) of low-altitude ozone concentrations at different latitudes and seasons. At high altitudes, the dominant form of odd oxygen is predicted by the models to be the oxygen atom. The dominant form of odd hydrogen near the ground is HO_2 , not unlike the troposphere of the earth. Still to be investigated are the ways in which Martian dust influences the chemistry of the Martian atmosphere.

The Jovian Planets

D. F. Strobel (Naval Research Laboratory) discussed the atmospheres of the Jovian planets, with special emphasis on Jupiter. The thermal structures of the Jovian planets (Figure 7) differ principally in the temperature offset owing to solar distance. Their composition is dominated by H_2 and He. Jupiter and Saturn have recently been the subjects of Voyager flyby analyses; combined with ground-based observations, these indicate the presence of CH_4 , C_2H_2 , C_2H_6 , NH_3 , and PH_3 in the atmospheres of both planets. In addition, CO has been detected in the Jovian atmosphere.

The observations can be compared with theory by using the deduced thermal structure, assuming thorough atmospheric mixing to great depths, and invoking chemical reactions that link the observed species. For Jupiter, this is done by starting with a parcel containing a variety of volatile constituents at an interior reference level of $p = 200$ mbar, $T = 2000^\circ\text{K}$, allowing the parcel to expand adiabatically, and calculating the levels at which the various liquid and solid phases condense out. The condensed species are assumed to remain as aerosols at these levels. Dense water clouds are calculated to form at 270°K , 80 km above the "surface" reference level where $p = 20$ mbar. Near the 200°K (90 km) level, it is thought to react with NH_3 to form a cloud of solid NH_4SH particles. White crystals of ammonia precipitate out at 154°K ($p = 800$ mbar, $z = 120$ km) to produce the visible upper cloud layer; a result supported by lines characteristic of solid ammonia in Jupiter's emission spectra.

It appears that many of the Jovian trace molecules are created high in the atmosphere (where solar photons are readily available) and are mixed down into the lower atmosphere. The color of the Great Red Spot and of some of the clouds is presumably a result of photochemical chemistry, although compounds of sulfur have also been suggested. Analyses of the hydrocarbon chemistry are complicated by two factors: uncertainty in the photochemistry of C_2H_2 and

C_2H_4 and the possibility of chemical effects from precipitating magnetospheric particles. Ongoing analyses of the Voyager data may place constraints on some of these processes.

The photochemistry of the atmospheres of Saturn, Uranus, and Neptune have received much less attention than has that of Jupiter and were not explicitly discussed at the symposium. There is every indication that similar approaches can be used to describe the chemistry of all the Jovian planets, however. All contain small hydrocarbons and probably ammonia. Spurred on by the wealth of Voyager data, they will no doubt be actively modeled over the next several years.

Titan, Triton, and Pluto

Little has been known of the atmospheres of these three bodies, and they have received little study as a result. The paucity of information was dramatically reversed for Titan by the Voyager flyby in November 1980. Previously thought to contain mostly methane, the atmosphere was found to be almost entirely N_2 . Trace amounts of CH_4 , C_2H_2 , C_2H_4 , C_2H_6 , and HCN were also detected by Voyager I, proving conclusively the existence of hydrocarbon photochemistry on Titan. The surface temperature and pressure, atmospheric scale height, and the presence of a 75°K temperature minimum at ~ 50 km were also established. Theoretical studies aimed at matching these observations and thus establishing the atmospheric chemical mechanisms are being vigorously pursued.

Triton and Pluto are known to possess frozen methane on their surfaces. At surface temperatures of 60 – 70°K , the methane vapor pressure from this methane ice will provide tenuous atmospheres. It is reasonable to suppose that the unobserved gases N_2 , H_2 , and Ar will be present as well. Until more detailed experimental observations can be obtained, little information will be forthcoming on the atmospheric chemistries of Triton and Pluto.

Summary

The photochemistry of planetary atmospheres has come of age in the last decade. From a single example, the modern atmosphere of the earth, the vigorous programs of planetary exploration have provided data on the atmospheres of Venus, Mars, Jupiter and Io, and Saturn and Titan. In addition, studies of geological records have provided substantial inferential information on the ancient atmosphere of the earth. Although we can still look forward to the observations of Voyager 2 of Uranus, Neptune, and Triton, eight examples of planetary atmospheres are available to the theoretical photochemist. Their similarities and their differences provide striking examples of the concomitant diversity and scientific rigor of nature.

Acknowledgments

The symposium on which this report is largely based was ably chaired by R. J. McNair, Menager of

News

Volcano Organization Formed

The past decade was an unusually eventful one for volcanology, with the 1973 eruption on Helmaey, Iceland, the 1975–1977 rumblings of the Soufriere Volcano on Guadeloupe, the huffing and puffing of Iceland's Krafla Volcano in 1975, the many eruptions of Mount Etna in Sicily in the 1970's, and the reawakening of Mount St. Helens in 1980 in the United States. In addition to their scientific duties, volcanologists have had to play an important role as advisors to administrators. For example, the political decision to evacuate Guadeloupe exposed French volcanologists to scrutiny more severe than previously experienced by members of the profession. These volcanologists met the challenge by completely reorganizing their volcano observations and by increasing their volcano research. In recognition of this recent reorganization and the associated modernization of volcano observatories on Guadeloupe and Martinique, the French government invited representatives from the world's volcano observatories and institutions to meet. The meeting, from February 18–21, 1981, resulted in the establishment of the World Organization of Volcano Observatories (WOVO). As its name implies, WOVO is concerned exclusively with volcano observatories and volcano monitoring; its activities will not duplicate the functions of existing international organizations, such as the International Association of Volcanology and Chemistry of the Earth's Interior (IAVCEI), which focus generally on volcanology and allied topics.

The principal objectives of WOVO are to create or improve ties between observatories and institutions directly involved in volcano monitoring; to facilitate the exchange of views and experiences by convening periodic, perhaps annual, meetings; to maintain an up-to-date inventory of instrumentation and manpower, which could be made available to any of the member institutions if a situation arises; to require scientific reinforcement; and to promote funding from international organizations, which could help to defray travel and related expenses of scientific reinforcement teams.


the Air Quality Program at NASA Headquarters. We thank him for his efforts. Our thanks also go to the symposium participants, both for their careful preparation and presentation and for their comments on earlier version of this report.



John F. Dewey is a Senior Research Scientist in the Atmospheric Environmental Sciences Division at the NASA Langley Research Center, Hampton, Virginia. He received a B.S. (Physics) from Brooklyn College, a M.S. (Meteorology) from New York University, a M.S. (Aeronomy and Planetary Atmospheres) and Ph.D. (Atmospheric Sciences), both from the University of Michigan. His current research interests are the origin and chemical evolution of atmospheres and the photochemistry of the troposphere, particularly assessing the impact of anthropogenic activities. He is the Principal Investigator of the Atmospheric Chemistry Experiment (ACE) aboard the NASA F-105B Storm Hazards Project aircraft, which recently obtained samples of air prior to, and during, a series of flights through thunderstorm lightning (the aircraft was struck by lightning several times). He is involved in the formulation and implementation of NASA's programs in both tropospheric and atmospheric evolution research. A member of the AGU since 1960, he serves on the Geophysical Monograph Board.



Thomas E. Graedel is a Member of Technical Staff in the Chemical Kinetics Research Department, Bell Laboratories, Murray Hill, New Jersey. He holds a B.S. (Chemical Engineering) from Washington State University, a M.A. (Physics) from Kent State University, and M.S. and Ph.D. degrees (Astronomy) from the University of Michigan. His current research interests are in computer modeling of the chemistry of the earth's troposphere and in experimental and theoretical investigations of atmospheric corrosion of metals and alloys. A member of AGU since 1965, he has served as Chairman of the Geophysical Monograph Board and is presently the Chairman of the Book Board and a member of the Publications Committee.



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ards warnings. The adequate minimum systems should include a seismic network of three stations (short-period, vertical seismometers and associated telemetry and recording systems) to locate the loci of seismic activity, a simple tremor or seismic alarm system to alert staff at night, and a continuously recording tiltmeter with associated radio-link telemetry. For calibration, the system also should have at least one non-electronic tilt-measurement array. Adequate minimum staff was defined by the group to be two geoscientists, one specializing in geophysics seismology and another in geology geodesy; one electronics engineer or technician; and two general purpose, physical science technicians. The group also deemed desirable, to augment the minimum monitoring, periodic occupations of a network of "dry tilt" stations, a modest triaxial (geodimeter) network, and a few leveling lines; observations of physical changes in the volcano's vicinity; collection of information resources, documentation, and manuals for instruments and monitoring techniques; and study of historical records to ascertain eruptive patterns.

Gudmundur E. Sigvaldason of Iceland was elected president of WOVO's executive board; other members are Tilling, Yokoyama, and J. L. Le Mouél (France).

For additional information on WOVO, contact Sigvaldason, Nordic Volcanologist Institute, Geosciences Building, University of Iceland, Reykjavik, Iceland.

This news item was contributed by G. E. Sigvaldason and is a compilation of reports by L. A. Mendes Victor of Portugal and Robert Tilling of the U.S. Geological Survey.

Luck + Merit = Grant

Granting approval to proposals submitted to the National Science Foundation (NSF) depends about as much on luck as it does on the scientific merit of the proposal, according to a study by the National Academy of Sciences' Committee on Science and Public Policy (COSPUP). In addition, concealing the names of the authors of the proposals, a practice known as "blinding," would not significantly change the outcome of the grant-awarding process.

"The fate of a particular grant application is roughly half determined by the characteristics of the proposal and the principal investigator, and about half by apparently random elements which might be characterized as the 'luck of the reviewer draw,'" according to Stephen Cole, Jonathan R. Cole, and Gary A. Simon in the COSPUP report summary, published in the November 20 *Science*. If researchers' proposals for NSF grants were rated again by an equally qualified group of reviewers, between 25% and 30% of NSF funding decisions could be reversed, they said.

However, "this should not be interpreted as meaning either that the entire process is random or that each individual reviewer is evaluating the proposal in a random way," they continued. To clarify the way in which the luck of the draw works, the report's authors suggest that the sources of reviewer disagreement be scrutinized.

"The great bulk of reviewer disagreement observed is probably a result of real and legitimate differences of opinion among experts about what good science is or should be. . . . As long as substantial reviewer disagreement, whatever its source, exists the fate of a particular proposal will depend heavily upon which reviewers happen to be selected," state Cole, Cole, and Simon.

COSPUP also found that "reviewers at major institutions did not treat proposals from applicants more favorably than did reviewers from lesser institutions." They found, in fact, the opposite trend. Length of career had no strong effect on the probability of receiving a grant. In addition, there was

Lindsay's objective is to bridge the gap between soil science and chemistry and to show that most reactions taking place in soils can be understood and predicted from basic chemical relationships. To achieve this objective he has written a book that simply describes the construction of nearly 100 mineral solubility diagrams and several aqueous distribution diagrams whose relevance to real soils is left as an exercise for the reader (both literally and figuratively). If you are looking for anything else that might tell under this title then you won't find it. Adsorption, colloid processes, ion exchange, and other surface effects are not considered.

Following two introductory chapters on aqueous chemical equilibria, each chapter is assigned to an element or constituent type, such as aluminosilicate minerals (chap. 5), carbonate equilibria (chap. 6), phosphates (chap. 12), chelate equilibria (chap. 18), and organic transformations (chap. 23). The other element chapters are all organized in a similar fashion: solubility of oxides and hydroxides, silicate solubility, hydrolysis, complexes and redox for aluminum (chap. 3), silica (chap. 4), calcium (chap. 7), magnesium (chap. 8), sodium and potassium (chap. 9), iron (chap. 10), manganese (chap. 11), zinc (chap. 13), copper (chap. 14), nitrogen (chap. 16), sulfur (chap. 17), silver (chap. 18), cadmium (chap. 19), lead (chap. 20), mercury (chap. 21), and molybdenum (chap. 22).

Although this book's central theme is the diagenetic representation of soil mineral solubilities, the classical use of solubility as the sum of all species is never represented. For example, the well-known parabolic pH-dependent curve for gibbsite solubility never appears. Instead, gibbsite solubility is shown as a linear pH function in terms of free aluminum ion only. This simplified approach is consistently used throughout the book.

The old and useless concept of α (circa 1923) has been born again in a single-term expression called $pe + pH$ by Lindsey to define the redox state of soil systems. This expression assumes that only one hydrogen ion is stoichiometrically associated with one electron per redox reaction, and any additional hydrogens are independent acid-base reactions. There is no demonstrable basis for this arbitrary rearrangement of the Nernst equation, and it leads to the confusing situation of treating pH as both a constant and a variable in the same reaction. This occurs in the chapter on sulfur, for example. It is unfortunate that $pe + pH$ has been

developed as a main feature of the book, since it is neither new nor useful.

The solubility, redox, and distribution diagrams are all based on a very extensive tabulation of thermodynamic data compiled by M. Sadiq and W. Lindsey. This carefully compiled reference is a valuable asset of the book except for several errors occurring in it, such as (1) some free-energy values for neutral aqueous species turn out to be the values for the fully dissociated species rather than the ion pair, triplet, etc.; (2) goethite is taken to be more stable than hematite, contrary to both field observation and reliable solubility studies; (3) several inorganic compounds that are not known to occur as minerals and have no known importance in soil chemistry are included in the solubility diagrams; and (4) the thermodynamic values are not critically selected or fully evaluated for several systems of interest to the reviewer. Nevertheless, the thermodynamic data are up to date (as of April 1978), extensive, and fully documented.

The strength of this book is the straightforward derivation of solubility diagrams from thermochemical data, and as such it would be useful as a teaching reference. The weakness of the book is the lack of real world applications and relevance. Students, for example, might be led to believe that minerals such as fayalite, wustite, forsterite, anorthite, and enstatite might be realistic solubility controls on soil water chemistry. Also the full complexity of heterogeneous soil systems with multiple simultaneous equilibria is not presented, and the effects of surface chemistry are absent. As a course textbook it has many shortcomings; but as a reference for teachers and researchers who are investigating mineral solubilities, it can be useful provided that the data base and the solubility diagrams are not accepted uncritically.

D. Kirk Nordstrom is with the Water Resources Division of the U.S. Geological Survey in Menlo Park, Calif.

Classified

EOS offers classified space for Positions Available, Positions Wanted, and Services. Rates per line: 1-5 lines—\$1.00, 6-11 lines—\$0.75, 12-26 lines—\$0.55.

POSITIONS AVAILABLE
Rates per line: 1-5 lines—\$2.00, 6-11 lines—\$1.60, 12-26 lines—\$1.40.

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Rates per line: 1-5 lines—\$2.50, 6-11 lines—\$1.95, 12-26 lines—\$1.75.

STUDENT OPPORTUNITIES
For special rates, query Robin Little, 800-424-2486.

POSITIONS AVAILABLE

University of Utah: Faculty Positions. The Department of Geology and Geophysics invites applications for four tenure track positions at the assistant to associate professor level.

1. **Economic Geology.** The specific area of expertise is open, however, preference will be given to candidates whose research interests are in geological, geochemical, or petrological characteristics of metallic mineral deposits.
2. **Sedimentary Geology.** Applicants should have research interests in modern or ancient sedimentary basins.
3. **Seismology.** Applicants with backgrounds and specialties in seismic reflection, seismic imaging or theoretical seismology will be given preference.
4. **Potential fields.** Geophysical work with specialty in potential theory including gravity and magnetism. (The closing date for this position is January 31, 1982).

A Ph.D. or equivalent is required. The vacancies are to be filled by September 1982; the closing date for applications for positions 1-3 is December 31, 1981. Applicants should submit a vita, transcripts, a letter describing his/her research teaching goals, and names of five persons for reference to William P. Nash, Chairman, Department of Geology and Geophysics, University of Utah, Salt Lake City, Utah 84112.

The University of Utah is an equal opportunity/affirmative action employer.

Faculty Positions. Two Faculty Positions in Geology. Tenure track positions in geology, assistant professorships. Ph.D. preferred or equivalent experience. Fall 1982.

Petrologist-Mineralogist. Candidate must be able to teach introductory geology, mineralogy, petrology, geochemistry, and optical mineralogy-petrology.

Invertebrate Paleontologist-Soft-Rock Geologist. Candidate must be able to teach courses in invertebrate paleontology, micropaleontology, sedimentation, and historical geology. Additional experience in recent marine environments highly desirable.

Applicants are expected to do research in their areas of expertise, and to lead students' field trips. Strong teaching and research commitments expected. Submit applications with resume and copies of transcripts, and have three letters of recommendation sent to the Chairperson, Department of Earth & Space Sciences, Indiana University-Purdue University at Fort Wayne, Fort Wayne, Indiana 46805. Indiana University-Purdue University is an equal opportunity/affirmative action employer.

Research Associate/Theoretical Physical Oceanography. Applications are invited for two postdoctoral research associate positions in the School of Oceanography, Oregon State University. Applicant will conduct research in theoretical modeling and observational comparisons of ocean circulation. Ph.D. in mathematics or the physical sciences. Submit resume, brief statement of research interests and three references to: Dr. Peern P. Niiler, School of Oceanography, Oregon State University, Corvallis, Oregon 97331. Closing date extended to 30 January 1982.

An Affirmative Action/Equal Opportunity Employer.

Positions in Oceanography/VIMS. The Virginia Institute of Marine Science (VIMS) School of Marine Science invites applications for two state funded, oceanography research and teaching positions at the levels of Senior Marine Scientist, VIMS is a broad-based marine science establishment with a mission to provide sound and timely advice to executive agencies and the legislature and to conduct intensive research programs. The School of Marine Science offers M.A. and Ph.D. programs with a faculty of 86 and 136 graduate students.

HEAD, DEPARTMENT OF GEOLOGICAL OCEANOGRAPHY (#113)
Applicants are sought with research interests in estuarine sedimentary geochemistry, dynamics of cohesive sediment transport, or estuarine and coastal geomorphodynamics. For further information contact Dr. Robert Byrne (VIMS), 804-642-2111 (Ext. 173).

ESTUARINE AND COASTAL HYDRODYNAMICS (Position #204)
A physical oceanographer with a strong interest in interdisciplinary approaches to complex estuarine and continental shelf problems is desired. For further information contact Dr. Bruce Nielsen (VIMS), 804-642-6131 (Ext. 234).

Candidates for both positions should have established research credentials and be dedicated to furthering the research and educational programs of the Institute. Demonstrated ability to generate extramural support is expected. Salary range is \$24,972 to \$34,107 and faculty rank is commensurate with qualifications. Applicants should send a comprehensive curriculum vitae, reprints, and at least three letters of recommendation by February 1, 1982, relating specific position of interest, to: Employment Manager, Personnel Office, College of Wm & Mary, Williamsburg, VA 23185.

An equal opportunity/affirmative action employer.

Physical Oceanographer. Royal Roads Military College expects to have a tenure track vacancy in Department of Physics effective 1 July 1982. Candidates should hold doctorate or near doctorate in physical oceanography preferably with experience in digital hardware and microcomputer applications. Applicant expected to be made an assistant professor level but salary and rank dependent on qualifications and experience. Relocation expenses can be provided. Outline include undergraduate teaching in physics and physical oceanography, and research in marine science. Applications should include complete dossier and names of three references and be sent to: Dr. E. B. Graham, Principal, Royal Roads Military College, FMO Victoria, B.C. V0S 1B0.

This competition is open to both men and women. Knowledge of English only is required. Only Canadian citizens or landed immigrants need apply. Tourist information relative to co-concours est deposable au troncet et peut être obtenue au Service à Dr. Graham.

An equal opportunity/affirmative action employer.

POSTDOCTORAL POSITION IN MARINE CHEMISTRY

Woods Hole Oceanographic Institution invites applications for the position of Postdoctoral Investigator. This position is a full-time position with basic research on the chemistry of the particle flux in the ocean and on the chemistry of sediment-seawater interactions, with particular emphasis on the transport of trace metals and trace nutrients. Preference will be given to applicants with training in geochemistry, trace element analysis, surface chemistry, or geochemical modeling. Send resume and names of three references to: Personnel Manager, Box 54P.

WOODS HOLE OCEANOGRAPHIC INSTITUTION

Woods Hole, MA 02543
An equal opportunity employer M/F/H

Planetary Scientist. SUNY Stony Brook, The Department of Earth & Space Sciences, anticipates that a tenure track faculty position may become available for a Ph.D. Planetary Scientist. Planetary Science within the department spans the range of planetary atmospheres, cosmochemistry, and planetary geophysics. The candidate should have a demonstrated record of accomplishment. The applicant is expected to pursue an active research effort and will be responsible for teaching courses at the undergraduate and graduate level.

Send vita and names of 3 references to: Dr. Roger Knacke, Dept. of Earth & Space Sciences, SUNY Stony Brook, Stony Brook, NY 11794.

SUNY Stony Brook is an EO/AA employer. AK 4226A.

Surficial Geology/Ground Water. Utah State University, Tenure track position starting spring quarter of 1982 or fall quarter of 1982. Ph.D. required. Rank and salary negotiable. Surficial geology with emphasis on geologic field studies and ground water with attention to both geologic and hydrogeologic aspects. Emphasis on the arid West. Closing date November 30, 1981. USU is an affirmative action equal opportunity employer. Department of Geology (07), Utah State University, Logan, Utah 84322.

Oceanographic Modeler. Ocean Data Systems, Inc. is seeking an applications oriented scientist to develop adaptive wave, HN and oil spill models for application in the Middle East on large CYBER computers. Position is in Monterey, experience in commercial applications of varied oceanographic models would be advantageous. Salary commensurate with ability and experience. Liberal benefits. Send resume to: M. C. R. Ward, Ocean Data Systems, Inc., 2400 Garden Road, Monterey, CA 93940.

Iowa State University of Science and Technology/Department of Earth Sciences.

Applications are invited for two tenure track faculty positions. The rank for each is at the assistant or associate professor level, dependent upon qualifications. The successful applicants will be expected to develop strong research and graduate student programs. Teaching duties will include undergraduate and graduate courses in the areas of sedimentary geology and geophysics. The area in mineral resources/economic geology. An applied field orientation is preferred. Iowa State has established a Mining and Mineral Resources Research Institute and an interdisciplinary minor in mineral resources. In order to support and develop these programs, we are seeking a person with research and teaching interests in this area. In addition to the appointment in the Department of Earth Sciences there will be full opportunities to interact with these programs.

Geomorphology. The second position is in the general field of geomorphology. Additional expertise in an area related to geomorphology, such as glaciology, engineering geology or remote sensing is also desired. A person with an applied field orientation is being sought.

Each appointment will be on an academic year basis. Opportunities are available for summer teaching appointments. Salaries will be commensurate with qualifications. Application deadlines for both positions are February 15, 1982; later applications will be accepted if a position is not filled. Positions are both currently available and are expected to be filled no later than fall, 1982. For application information please write to:

Bert E. Nordlie
Department of Earth Sciences
253 Science I
Iowa State University
Ames, Iowa 50011

Iowa State University is an equal opportunity/affirmative action employer.

South Dakota School of Mines and Technology. Applications are invited for two positions in the Department of Geological Engineering. Both involve teaching of undergraduate and graduate levels, these direction and development of research.

Geological Engineering: specialty in rock or soil mechanics, site evaluation, geohydrology, petroleum engineering or engineering seismology. Industrial experience helpful. A Ph.D. in some area of engineering is preferred. One Deposit: area of specialization is open. The successful applicant will work closely with the newly established Institute for the Study of Mineral Deposits. The Ph.D. is required.

The Department has an undergraduate enrollment of 100 majors and a graduate enrollment of 50. Field experience in geology and engineering are emphasized. Interested persons should send a resume and three letters of recommendation to: Alvin J. Lafferty, Dept. of Geological Engineering, South Dakota School of Mines and Technology, Rapid City, SD 57701.

SDSMT is an equal opportunity/affirmative action employer.

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Oceanographer. G6-1360-12, Salary \$28,245-\$38,723. The Remote Sensing Branch of the Naval Ocean Research and Development Activity (NORDA) is seeking qualified applicants for the position of Oceanographer. Duties include: Serving as principal investigator for planning and organizing basic and applied scientific investigations of radio probing of the ocean surface, and interpreting the results of these investigations in terms of oceanographic parameters. Specific areas of investigation will include the detection and analysis of ocean fronts and eddies through the use of satellite-borne altimeters. Responding to Announcement No. 01-039, send a current SF-171 no later than 21 December 1981 to the Civilian Personnel Office (Code 140A), Naval Ocean Research and Development Activity, NSTL Station, MS 36558 or call 601-858-4841 for appropriate forms or additional information.

Send vita and names of 3 references to: Dr. Roger Knacke, Dept. of Earth & Space Sciences, SUNY Stony Brook, Stony Brook, NY 11794.

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Surficial Geology/Ground Water. Utah State University, Tenure track position starting spring quarter of 1982 or fall quarter of 1982. Ph.D. required. Rank and salary negotiable. Surficial geology with emphasis on geologic field studies and ground water with attention to both geologic and hydrogeologic aspects. Emphasis on the arid West. Closing date November 30, 1981. USU is an affirmative action equal opportunity employer. Department of Geology (07), Utah State University, Logan, Utah 84322.

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Research Associate/Theoretical Physical Oceanography. Applications are invited for two postdoctoral research associate positions in the School of Oceanography, Oregon State University. Applicant will conduct research in theoretical modeling and observational comparisons of ocean circulation. Ph.D. in mathematics or the physical sciences. Submit resume, brief statement of research interests and three references by 1 January 1982 to: Prof. Peern P. Niiler, School of Oceanography, Oregon State University, Corvallis, Oregon 97331. An affirmative action/equal opportunity employer.

Faculty Position/AGU. The Department of Chemistry at Arizona State University invites applications for a possible tenure track position at the assistant professor level in one of the following areas: (1) Synthetic Solid State Chemistry; (2) Surface Chemistry; and (3) Atmospheric or Low-Temperature Geochemistry. Candidates should have demonstrated ability to develop a vigorous and innovative research program in one or more of the above areas and have a commitment to instructional excellence. A resume, brief description of research plans, and three letters of recommendation should be sent to Professor William S. Oleson, Chair, Search Committee, Department of Chemistry, Arizona State University, Tempe, Arizona 85287. EOAA employer.

Structural Geology/University of Wyoming. The University of Wyoming, Department of Geology and Geophysics seeks applications for a tenure track appointment in structural geology expected to be available beginning fall semester 1982 or earlier. Duties will include teaching of undergraduate and graduate courses in structural geology, supervising MS and Ph.D. theses, and research in structural geology. Appointment as assistant professor level is preferred, but applicants requesting appointment of higher rank will be considered. Salary open. Applicants must have Ph.D. degree and be versed in quantitative theory as well as field applications or modern structural geology and regional geology.

Applicants should provide, by January 1, 1982, a resume, three letters of reference, and a letter of application indicating a statement of current research interests and courses which the applicant feels qualified to teach. Applications should be sent to:

Dr. Robert S. Houston/Head
Department of Geology and Geophysics
University of Wyoming
Laramie, Wyoming 82071-3008.

The University of Wyoming is an equal opportunity/affirmative action employer.

Mineral University/Water Resources Program. Department of Civil Engineering, Department of Civil Engineering, University of Wyoming, Laramie, Wyoming 82071-3008.

The University of Wyoming is an equal opportunity/affirmative action employer.

Hydrology Tenure Track Position at Assistant or Associate Professor Level. Candidates should be a specialist in some quantitative aspect of hydrology with demonstrated skills in formulating hydrologic models, and a background in hydrologic phenomena. Academic or professional experience at Ph.D. level required. Starting date negotiable but could be as early as August 1982. Resumes, etc., should be received by March 1, 1982. Interested persons should request job description from: Dr. E. S. Simpson, Chairman, Search Committee, Department of Hydrology and Water Resources, University of Arizona, Tucson, Arizona 85721.

Send opportunity/affirmative action employer.

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Postdoctoral Fellowships/Department of Oceanography, University of British Columbia. Available January 1, 1982 for studies of the mineralogy and geochemistry of deep ocean ferromanganese nodules and the relationships between nodules and their associated sediments. Salary \$10,000. Send curriculum vitae, statement of research interests, and names of three references to: S. E. Calvert, Department of Oceanography, University of British Columbia, Vancouver, B.C., Canada V6T 1W5.

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Send opportunity/affirmative action employer.

Hydrology Tenure Track Position at Assistant or Associate Professor Level. Candidates should be a specialist in some quantitative aspect of hydrology with demonstrated skills in formulating hydrologic models, and a background in hydrologic phenomena. Academic or professional experience at Ph.D. level required. Starting date negotiable but could be as early as August 1982. Resumes, etc., should be received by March 1, 1982. Interested persons should request job description from: Dr. E. S. Simpson, Chairman, Search Committee, Department of Hydrology and Water Resources, University of Arizona, Tucson, Arizona 85721.

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Send opportunity/affirmative action employer.

Research Associate/Electron Microprobe. The Electron Microscopy Center at Texas A&M University invites application for the position of Electron Microprobe Specialist. Applicants should possess a working knowledge of WDS and EDS applications and accompanying computer and software programs and preferably have had experience in the Geological Sciences.

The primary duties of the position are to oversee and maintain (with the aid of service contracts) the electron microprobe and ancillary equipment and to assist in teaching graduate courses laboratories dealing specifically with electron microprobe analysis. Salary will be up to \$25,000/12 months.

Applicant should send supporting data and letters of recommendation to:

Dr. E. L. Thurston
Texas A&M University
Biological Sciences Building
College Station, Texas 77843

Texas A&M is an equal opportunity/affirmative action employer.

Structural Geology/University of Illinois at Champaign-Urbana. (Search reopened) The Geology Department is seeking a structural geologist for a tenure-track (assistant professor) faculty position. A Ph.D. is required. Salary open. The successful candidate will be expected to teach advanced undergraduate and graduate courses in structural geology and establish a research program. For equal consideration, applications, including the names of three referees, should be sent by February 1, 1982 to Dr. O. E. Anderson, Department of Geology, University of Illinois, 245 Natural History Building, 1301 West Green Street, Urbana, IL 61801-2986 (217) 333-5713

phen P. Huestis, R. H. Hunter, Liaquat Husain, Tissa Ilengasekara, Andrew P. Ingersoll, Dallas B. Jackson, Ian N. S. Jackson, Gary K. Jacobs, Ivan C. James II, J. Douglas James, Raymond Jeanloz, Ronald E. Johnson, Tracy L. Johnson, Blair F. Jones, Glyn M. Jones, Gieskes Jansin, Derrell L. Judge, Donna M. Judy.

Douglas L. Kane, Robert B. Kasper, Keith W. Keshava, Susumu Kato, George V. Keller, William W. Kellogg, B. L. N. Kennell, Iwane Kimura, Charles T. Kincaid, Fred W. Klein, Georges deVries Klein, Martin J. Koonen, A. Kroener, James L. Krumhansl, Roman Krzyzowski, Philip R. Kyle, John L. Le Breque, Gerard Lachapelle, L. Stephon Lau, Michael T. Ladbetter, Baum K. Lee, Tay-How Lee, Chin S. Lin, Donald H. Lindsay, John G. Liu, Richard Lively, Max Loewenstein, John M. Logan, Leland Timothy Long, Robert P. Lowell, Allen Lowrie, Charles A. Luther, Gregory A. Lyzenga.

William D. MacDonald, Peter K. MacKinnon, Syukuro Manabe, David H. Meiks, Bruce D. Marsh, James G. Marsh, Margerit Elizabeth Marshall, George B. Metanga, Barry H. Meuk, Barbara B. Mevko, John C. Maxwell, Ted A. Maxwell, Hens G. Mayr, Samuel W. McCandless, Jr., William R. McCann, James S. McClain, Kenneth A. McGee, L. D. McGinnis, Christopher P. McKay, Karen C. McNelly, A. G. McNamara, John K. McRaney, Harry Y. McSweeney, Jr., Judson Mead, W. D. Means, Scott G. Melford, R. Melzner, Robert L. Melvin, Michael Mendillo, James W. Mercer, Gerald Meyer, Jean-Francois Minster, Poive Mireille, Jerry L. Modisette, Harold O. Mofield, Volker A. Mohren, Erik Moll-Christensen, Judith B. Moody, Christopher N. K. Moers, Michael M. Mooradian, Stearns A. Morse, Paul Morgin, Helmut Moritz, Sharon Mooshor, Robert A. Muller, John J. Murray.

Kazuki Nakamura, Manuel Nathenson, Kenneth H. Nealson, Shlomo P. Noumen, Carl A. Newton, V. V. Nguyen, Yasunori Nishida, Clydo Nishimura, Chirios A. Nitrouer, James S. O'Brien, Richard J. O'Connell, Anne O'Donnell, Emile A. Okal, Julio C. Olimpio, John J. Oliveira, Peter Olson, Michael L. Oringio, Peter J. Palmadesso, Donald F. Palmer, Hons A. Panofsky, K. Papadopoulos, George K. Parks, Mervyn S. Peterson, Malcolm R. Patterson, William C. Patzer, Adolf K. Paul, Rudolf Penndorf, Henry T. Perkins, Mary Jane Perry, Mark E. Pesses, John A. Philpotts, Richard E. Pieper, Nicklas G. Pisias, Philip D. Rabinowitz, Chandra S. Rai, Sundar Rajon, W. T. Rawlins, Walter J. Rawls, David K. Roa, George Roid, Stephen P. Reidel, Patricia Hofer Roff, David W. Reinisch, Stuart R. Romz, Richard Reynolds, J. Michael Rhoads, Jack M. Rice, Paul G. Richards, John B. Robertson, Pierre-Yves F. Robin, John Rodgers, Alice M. Rogan, William M. Roggenhagen, John V. Ross, Kelly E. Rowe, John C. Rucklidge, John Rundle, Ernest Henry Rutter.

Neil E. Salisbury, Patricia F. Sailer, Charles G. Sammls, David Sandwell, Jorge L. Sarmiento, Tetsuya Sato, Y. Sato, Kim D. Saunders, Edward William Sawyer, Jean-Guy Schilling, Stephen H. Schneider, Jeffrey P. Schubert, Frederick Roy Schutt, James D. Schumacher, Margaret Ann Shea, Charles R. Shearer, Neil R. Sheeley, Jr., Alan Shiller, Kunihiko Shimazaki, Thomas H. Shipley, Peter N. Shive, Robert A. Shuchman, Kerry E. Steh, El A. Silver, David W. Simpson, Vijay P. Singh, A. K. Sinha, Ogbazghi Sum, William L. Sjogren, Allan Skorpens, Don F. Smart, Eugene Smart, Christopher R. Smith, Sheldon Sommer, Bengt Somnerup, D. J. Southwood, Warren T. Spaeth, Jr., P. Srinivasan, Robert J. Stamer, Charles H. Stayer, Heinz G. Stefan, Maureen Steiner, Yaron M. Sternberg, Edward F. Stoddard, Frank S. Stone, D. W. Strangway, Ronold Street, William D. Stuart, Brian W. Stump, Lynn R. Sykes.

Joern Thiede, Gary E. Thomas, George A. Thompson, Lonnie G. Thompson, John Thrallkill, John Toole, Robert J. Tracy, Georges Vachaud, Janet Van der Veen, Gilbert K. Van Dyke III, Mario S. Vassiliou, William W. Vaughan, James F. Vedder, Richard Carl Vistbuchen, Jr., N. J. Vlaar, Marc Vusgnat, Hare S. Wolf, R. J. Wagonet, Jean Jacques Wegner, Brian Wensmaker, Robert C. Ward, Richard Wall, Deborah J. Wechsler, Andrew J. Weinheimer, David B. Wenner, Frank J. Wentz, Daniel Wieder, Donald U. Wise, Peter M. Woiceshyn, Roger G. Wolff, David R. Wones, Eric Wood, Mark T. Woods, David A. Woolliar, Marinus J. R. Wortel, Peter J. Wyllie, William Yeh, S. H. Yungul, Savosh M. Zand, Alex Zaporozec, Vrdor Ziolknick, Ernst G. Zurlueh.

Candidates Statements: Section Presidents-Elect

In the next few issues, comments from all candidates for office of section president-elect will appear. The statements for Geodesy, Geomagnetism and Paleomagnetism, Hydrology, and Solar-Planetary Relationships sections appear below.

B. D. Tapley (Geodesy)

The following subjects are of importance to the AGU at large and the geodetic community in particular.

1. A significant effort should be made to improve communication between the organizations involved in geodetic operations, which include both industries and federal agencies, and research activities associated with the development and initial exploitation of contemporary geodetic instruments and theoretical approaches to the use of the new data types (VLBI, GRAV, Global Positioning System, satellite altimetry, GPS, etc.). Efforts to stimulate the dialogue between the research and application communities will bring new techniques into the operations area at the earliest feasible time and will provide a proper stimulus for the research and development activities. Such dialogue should be enhanced through special sessions at the biannual AGU meetings and by sponsoring intersociety workshops and specialized symposia.

- The AGU should initiate an effort to evaluate and improve the university educational offering in the area of geodesy. A geodesy section sponsored by the Education Affairs Committee should be given the responsibility for developing plans for improving the quality of the educational programs in geodesy and geophysics.
- The role of geodesy in other related areas should be identified and actions initiated to satisfy the geodetic requirements that emerge in these areas. Of particular significance are the roles that geodetic methods play in the areas of oceanography (definition of the geoid and gravity field and ocean surface and ocean bottom positioning) and seismology (precise point positioning, the time rates of change of point position, polar motion, etc.).
- The role of large data set collection, archiving, and dissemination are topics of major concern to the geodetic community. The Geodesy section activities, through the section meetings and/or technical committees, should seek to distribute information on the availability of data for geodetic studies, to define archival requirements for geodetic data base, and to identify requirements that are not being met by current agency actions.

If elected, I will work toward achieving the objectives outlined above.

Petr Vanicek (Geodesy)

North American geodesy has been suffering from a lack of identity. This conlition has seen some of the most spectacular geodetic achievements; yet, often enough these were born under the auspices of space science, tectonophysics, oceanography, etc. Some of my more pragmatic colleagues will argue that there is nothing wrong with this 'throw out the baby with the bathwater' approach, but I believe that geodesy would be better served if geodesy were, once more, recognized for what it is: a scientific discipline in its own right.

Geodesy is the discipline that concerns itself with the geometry of the earth and its gravity field. Including the temporal variation thereof. Thus geodesy is as much a part of geophysics as geometry is a part of physics. It is on these philosophical foundations that the Geodesy Section should cooperate with the other sections of AGU. It is in this position that the Geodesy Section should take when asking itself what is there that geodesy can do for the other disciplines and what is there that the other disciplines can do for geodesy. It is this point of view that I should adopt if elected to the presidency.

On a more down-to-earth level, the section may consider doing more for the promotion of AGU among professional colleagues and students alike. The promotion should include soliciting research papers as well as review papers for JGR and RGSP and soliciting manuscripts for publication in the AGU Monograph Series. As an example of other possibilities, the section may consider launching a campaign to evaluate geodetic content in various pertinent university programs on this continent and publish the results. To end, I should like to thank the Nominations Committee for the honor they bestowed upon me by nominating me, even though I am not a United States citizen.

Christopher G. A. Harrison (Geomagnetism and Paleomagnetism)

The main function of the American Geophysical Union is to provide communication channels between geophysicists. This is done by the meetings and the publications programs of AGU. Actions taken at council meetings can affect both of these programs, and one of the jobs of council members is to ensure that the various programs remain effective. This is especially true during times of rapid change, which we are seeing today. For instance, the publications program of AGU is undergoing several changes that may affect members affiliated with the GP section. The Red JGR has had a significant increase in size under the editorship of Tom Ahrens, and a new journal, *Tectonics*, has been started in collaboration with the European Geophysical Society. The annual meetings are also undergoing changes, with the western one, in particular, growing through the years. The increase in submitted abstracts may necessitate significant changes in the way the meetings are organized. In order that a council member may adequately represent his or her section members at council meetings, it is important that there be good communication between the council representative and the members of the section. The lunch meetings of the GP section, which take place at the annual meetings, provide some opportunity for exchange of ideas which I would like to see made greater use of. I shall also try and discuss issues with individuals during the annual meetings so as to be better prepared to represent GP at the council meetings.

Other things that are of importance to GP members are that magnetic observatories continue to operate, that new orbiting magnetic observatories be launched in a timely manner, and that our data bases at the National Data Center be kept up-to-date.

Neil D. Opdyke (Geomagnetism and Paleomagnetism)

It is a great honor to be selected by the nominating committee to run for president of the GP section. It is my opinion that the section is doing quite well; however, there is one matter that must be addressed if we are to remain a coherent active group. The problem to which I am referring is the increasing tendency for the sections to split into a West Coast and an East Coast group based upon the respective national meetings. How this tendency can be countered during a time of increasing difficulty in obtaining

additional funds for travel is difficult to see. A possible route may be to attempt to coordinate the symposium offered at the two meetings to have as wide an appeal as possible and to invite speakers to these symposium from both coasts. Therefore, I elected, I would attempt to be more involved in the coordination of the programs for the two meetings so as to maximize the interaction of what is rapidly becoming two constituencies.

R. Allan Freeze (Hydrology)

I believe that the Hydrology Section of AGU is currently in good hands, and I would not envisage any major shifts in direction should I be elected president of the section.

The primary obligations of the section's executive are to ensure efficiency and quality in the publications program and in conferences and symposia. On the publications side, our research journal, *Water Resources Research*, has a strong reputation, and I would provide the necessary support to the editors to ensure the continuation of this reputation.

One of the long-standing problems in the section is how to bridge the gap between the relatively small number of research-oriented hydrologists who tend to publish in *WRR* and the much larger number of practicing hydrologists who form the backbone of the section. The user-oriented publications of the *Water Resources Monograph Series* have the potential to be the primary vehicle for bridging the gap. I would like to see this program improved and expanded.

With respect to symposia, I would like to see continued expansion of the Chapman Conference program. All the Chapman Conferences to date have been successful and well attended. Recent annual meetings, despite excellent technical sessions, have experienced relatively poor attendance. This problem deserves continued attention.

I understand that the technical committee structure in the section is healthy and active. I hope to see the members of these committees play a leadership role in the development of Chapman Conferences and in the organization of technical sessions at the annual meetings. In particular, I would like to see more technical sessions that cross the interdisciplinary boundaries within hydrology.

On a broader front, I would like to see greater interaction between the Hydrology Section and the other sections of AGU. I think we should abandon our traditional role as outsiders and jump headlong and headstrong into AGU affairs, both political and scientific. In the long haul, I believe that this is the best way to project the interests of the Hydrology Section within AGU and throughout the broader scientific community.

George C. Reid (Solar-Planetary Relationships)

The discipline represented by the Solar-Planetary Relationships Section of AGU are facing a period of crisis brought about by the proposed major decreases in research funding. They are likely to be affected to an even greater degree than other areas of basic science, since they have become so dependent on spacecraft missions, whose future is, to say the least, unpromising. In this situation, a slight shift in the emphasis of AGU might be worth considering. Traditionally, AGU has not played an active 'lobbying' role, restricting itself mainly to the development of communication among scientists through the twin media of publications and meetings. This should certainly continue to be its principal function, but I should like to propose an increase in emphasis on communication between scientists and the outside world. The market for popular science is stronger than ever before among the educated public, and Solar-Planetary Relationships has an interesting story to tell. I feel that our section should work, through its officers, with the Public Affairs Committee of AGU to convey the inherently exciting aspects of our work to the general public and thereby elicit to build an effective lobbying force among the public, most of whom are not aware of the serious dangers that the future holds for science.

There are a few specific issues facing both the Union and the section that will demand attention in the near future. The proposal that the Aeronomy subsection be merged with the Meteorology Section to form an Atmospheric Science Section is probably the one of most direct and immediate concern. While this proposal has obvious merits, there are potential problems that need to be examined and laid to rest before a final decision is made. Some restructuring within the section may also become desirable, whether or not Aeronomy remains as part of SPR.

The AGU journals represent a valuable international resource and are among the world's leading journals in their fields. This is especially true of blue JGR, relative to the disciplines of SPR, and every effort should be made to maintain, or even to enhance, its position. I elected, I propose to work with the Editor, the Publications Committee, and the Council to support the cause of blue JGR and to avoid any changes that may be detrimental to its international standing. The problem of journal finances is likely to increase in importance in the next few years, and the long-standing question of the balance between individual and institutional subscription rates may need reexamination. In these issues, and in others as they arise, I shall try to act in the best interests of the section, of the Union, and of the discipline itself.

Christopher T. Russell (Solar-Planetary Relationships)

The primary function of the AGU is and should continue to be scientific communication. The AGU fosters this communication in many ways: through its weekly newspaper, *Eos*; through its research journals, review journals, and books; and through its sponsorship of annual meetings and Chapman Conferences. Three aspects of the journals program are of most concern to the membership: editorial quality of content, cost, and time to publication. I am quite

convinced that the editorial content of the AGU journals, in the SPR section the journals of choice are the JGR and GRL. However, I am not satisfied with the length of the time from submission to publication for JGR and to a lesser extent for GRL. The decline in subscriptions alarms me. The more AGU journals are in scientist's offices, the more the articles within them will be used. Perhaps subscription rates need to be decreased at the expense of page charges.

In the area of the annual meetings the AGU seems to be checking on its success. Is there a reasonable solution to the problem of overcrowded rooms and conflicting parallel sessions? I believe we should analyze carefully why this crowding has come about before offering solutions. I do not believe poster sessions are the instant cure-all to the problem, nor is the arbitrary rejection of abstracts desirable. I entered the field too short a time ago to believe the 'old boys' know what should be accepted. We must nurture new blood and innovation. The Chapman Conferences have proven themselves to be effective and provide an atmosphere unobtainable at the annual meetings. However, have we used them most effectively? Have the subdisciplines in SPR had their fair share of these conferences? I believe that the SPR section officers should take a more active role in promoting Chapman Conferences to cover the needs of the members.

We should not overlook the other roles of the AGU, even though they do not touch us as directly or as often as meetings. We must encourage good young people to enter the field, even with the uncertain job market. There will always be a need for good people in our field. At the present time the employment opportunities in the various sections of geophysics is uneven. The AGU can assist in smoothing this imbalance by matching prospective employees with available geophysicists when they find themselves outside

the 'old boys network' in a particular area. Finally, we live in changing times. The amount of money available for research seems to be going down. The stillness of the federal government regarding the need for basic research seems to be taking a turn for the worse. At a very minimum we need the AGU to act as a listening post for us and to keep the membership informed about the issues that affect geophysicists. However, perhaps the time has come when the AGU and its members should take a more active role in informing the general public and those in the federal government about what we do.

One last point that has concerned me over the years is the rate of errors in the membership service area: lost subscriptions, lost articles, and the like. There has been much improvement over the last year, but there is still room for further improvement, and as a council member I would continue to monitor this area and assist in promoting a mistake-resistant system at AGU headquarters.

BALLOT CHANGES

Withdrawn

Donald R. Nielson has withdrawn as a candidate for president of the Hydrology Section.

Petition Candidate

Joseph N. Barfield has been approved as an additional candidate for Secretary of Magnetospheric Physics for the Solar-Planetary Relationships Section. His biography and picture appear below.



Barfield

Age 40; he has been a member of the AGU since 1968. He is a staff scientist with Southwest Research Institute in San Antonio, Texas. His research interests include magnetospheric substorm dynamics, magnetospheric current systems, and ULF waves in the magnetosphere. He earned his B.A. in physics from Rice University in 1965 and his Ph.D. in planetary and space physics from UCLA in 1972. He was a research physicist with the NOAA Space Environment Laboratory from 1972 to 1979, where he participated in the analysis of Explorer 45 particle data and was responsible for processing of the NOAA magnetometer data bases. Instrumental in the planning of the North American IMS magnetometer data base, he was the IMS data coordinator for NOAA SEL. After 2 years as an independent consultant, he joined Southwest Research Institute, where he has continued his magnetospheric research and has participated in the design of the data processing system for the Dynamics Explorer space plasma data. He has authored or coauthored 28 scientific publications, including 17 in AGU publications. He has served as referee for various journals and as organizer for two international scientific conferences. He has also served as session chairman at several AGU meetings and as an invited speaker at several international symposia. Barfield is presently a member of the National Academy of Sciences Panel on the International Magnetospheric Study (IMAS) Phase II and is a member of a number of working groups in space physics.

Meetings

Aerospace Sciences Meeting

The American Institute of Aeronautics and Astronautics' 20th Aerospace Sciences Meeting, to be held in Orlando, Fla., on January 11-14, 1982, will include four main sessions and a special presentation by Esler Davis, Voyager project manager, entitled 'Voyager: A Parade of Giants.'

Results from the Spacecraft Challenging at High Altitude (SCATHA) program will be summarized in one session. For more information, contact A. L. Vampola, The Aerospace Corp., P.O. Box 92957, Los Angeles, CA 90009. Seven papers will describe the science and engineering of a proposed space mission in a session entitled 'Star Probe: A Mission to the Sun.' For additional information, contact D. Sonnabend, Jet Propulsion Laboratory, Pasadena, CA 91103. 'Active Space Experiments' will include six papers on the results of active stimulation of space plasmas. Details can be obtained from S. Kaye, Plasma Physics Laboratory, Princeton University, Princeton, NJ 08544. Six papers will evaluate the effects of the environment of space on spacecraft systems. For additional information, contact C. P. Pike, Air Force Geophysics Laboratory, Hanscom AFB, MA 01731.

General information about the meeting can be obtained from Paul F. Mizera, The Aerospace Corp., P.O. Box 92957, Los Angeles, CA 90009 (telephone: 213-648-8514).

Water: Indiana's Abundant Resource

A call for papers has been issued for the Third Annual Indiana Water Resources Symposium, sponsored by the Indiana Water Resources Association. Papers are invited on all aspects of water resources, but special emphasis will be placed on papers addressing irrigation, urban hydrology, water quality, groundwater hydrology, lakes and wetlands

hydrology, hazardous wastes, or multipurpose water uses.

For a paper to be considered for the meeting, which is scheduled for June 9-11, 1982, in South Bend, Ind., authors should submit an original and four copies of detailed abstracts of their papers no later than January 8 to John E. Fisher, Chairman, Third Annual Indiana Water Resources Symposium, Lawson-Fisher Associates, 525 West Washington St., South Bend, IN 46601 (telephone: 219-234-3167). Abstracts should not exceed 250 words end area to include the paper title, author name(s), affiliation(s), address(es), and telephone number(s). The senior author should be noted with an asterisk. Authors will be notified at the end of January. Camera-ready copy must be submitted by April 1.

Organizations wishing to present an exhibit at the symposium also should contact Fisher by January 8.

Paleoenvironment of East Asia

The Centre of Asian Studies at the University of Hong Kong is proposing to convene a conference in early December 1982 on the Paleoenvironment of East Asia from the Mid-Tertiary. The conference objective will be to review the evidence for the sequence of geology and paleoclimatology that parallel evolution of biological ecosystems in Asia. This review should lead to a definition of the nature of the environment in which hominids and early forms of *Homo* evolved in East Asia, according to the centre.

Geologists, geomorphologists, paleoceanographers, paleontologists, paleoclimatologists, paleoecologists, paleozoologists, and paleoanthropologists are expected to contribute to the conference. In addition, specialists from China, Japan, Korea, and Taiwan are expected to describe their advances in these disciplines.

Geophysical Year New Listings

(Italics indicate meetings sponsored or cosponsored by AGU.)

1981

Dec. 18-19 Annual International Meeting of the Working Group on Mediterranean Ophiolites, Florence, Italy. (Luigi Boccalone, Istituto di Petrografia, Via Gramsci 6, 48100 Parma, Italy.)

1982

Jan. 3-9 Annual Meeting of the American Association for the Advancement of Science, Washington, D.C. (AAAS Meetings Office, 1776 Massachusetts Avenue, N.W., Washington, DC 20036.)
Jan. 11-14 Symposium on the Understanding of Hydrologic Processes at the Basin Scale, Caracas, Venezuela. Sponsors, Universidad Simón Bolívar, UAB, (Ignacio Rodríguez-Luque, Universidad Simón Bolívar, Apartado Postal 80 689, Caracas 1061, Venezuela.)
Jan. 13-15 National Radio Science Meeting, Boulder, Colo. Sponsors, U.S. National Committee for the International Union of Radio Science, IEEE (U.S. National Committee for URSI, National Research Council, 2101 Constitution Avenue, N.W., Washington, DC 20418.)
Jan. 24-29 Conference on Origins of Plateaus and Related Fields in the Magnetosphere, Yosemite National Park, Calif. Sponsors, NASA, AGU, (F. T. Barker, Center for Solar-Wind and Space Sciences, California State Univ., Fullerton, CA 92632.)
Jan. 25-29 Fourth Annual Conference on the Solar Wind and Space Sciences, Greenbelt, Md. (P. T. Taylor, Code 922, NASA/Goddard Space Flight Center, Greenbelt, MD 20771.)
Feb. 5-12 Third International Geodetic Symposium on Satellite Doppler Positioning, Las Cruces, N. Mex. Sponsors, Defense Mapping Agency, National Ocean Survey, AGU, (Richard Paul, Defense Mapping Agency, Hydrographic/Topographic Center, 6500 Brooks Lane, N.W., Washington, DC 20315.)
Feb. 19-18 Ocean Sciences AGU/ARL (American Research Society of Limnology and Oceanography) Joint Meeting, San Antonio, Tex. (Meetings, AGU, 2000 Florida Ave., N.W., Washington, DC 20006.)
Feb. 25-28 13th Annual Meeting of the International Erosion Control Association, Salt Lake City, Utah. (M. McMillan, Erosion Control Consultants, P.O. Box 135, Provo, UT 84604.)
Mar. 22-24 International Symposium on Hydrothermal Reactions, Yokohama, Japan. Sponsors, Tokyo Institute of Technology, (Shigeyuki Somya, Research Laboratory of Engineering Materials, Tokyo Institute of Technology, Nagatsuta, Midori, Yokohama, 227 Japan.)
Mar. 24-27 Conference on Earthquake Hazards in the Eastern San Francisco Bay Area, Hayward, Calif. Sponsors, USGS, East Bay Council on Surveying and Mapping, Calif. Div. of Mines and Geology, Woodward-Clyde Consultants, Calif. and Geod. (John Hirschfeld, Dept. of Geological Sciences, California State University, Hayward, CA 94542.)
Apr. 11-18 Penrose Conference on Antarctica, Shantou National Park, Va. Sponsors, GSA, (Jan W. D. Dettl, Leont-Joherty Geological Observatory, Columbia University, Palisades, NY 10964.)
Apr. 19-21 Cordilleran Section, Geological Society of America and Geological Society of America American Annual Meeting, Anaheim, Calif. (Neil Maloney, Earth Science Department, California State Univ., Fullerton, CA 92632.)
Apr. 27-29 Chapman Conference on Rainfall Retention, Urbana, Ill. (Meetings, AGU, 2000 Florida Ave., N.W., Washington, DC 20009.)
May 3-5 Chapman Conference on the Dis-

continuation in Rock Their Role and Significance in Geologic Processes, Santa Fe, N. Mex. (Meetings, AGU, 2000 Florida Ave., N.W., Washington, DC 20009.)
May 5-7 14th International Liège Colloquium on Hydrodynamics of Ecosystems, Liège, Belgium. Sponsors, IAPSO, UNESCO Marine Sciences Division, EGS, International Oceanographic Association, AGU, (Jacques C. J. Nihou, University of Liège, Mécanique des Fluides Géophysiques-Environnement, 28-Sart Tilman, 5-4000 Liège, Belgium.)
May 7-20 General Meeting of IAG, Tokyo, Japan. (N. Nakagawa, Geophysical Institute, Kyoto University, Sakyo-ku, Kyoto 606 Japan.)
May 10-12 Annual Meeting of the Canadian Geophysical Union, Downsview, Ontario, Canada. (O. E. Smylie, Department of Physics, York University, Toronto, Ontario, Canada M3J 1P3.)
May 12-14 Fourth International Conference on Planning and Management of Water Resources for Industrial, Agricultural, and Urban Use, Marseille, France. Sponsors, Commission Européenne Méditerranéenne de Planification des Eaux (C.E.M.P.E.), Société des Eaux de Marseille (S.E.M.), le Bureau de Recherches Géologiques (B.R.G.), le Bureau des Recherches Géologiques et Minières (B.R.G.M.), Centre de Formation Internationale à la Gestion des Ressources en Eau (CEIFORE), UNESCO, Commission des Communautés Européennes, Association des Hydrologues (A.H.). (Secrétariat de la Conférence, Société des Eaux de Marseille, 25 rue Edouard Delacroix, 13008 Marseille, France.)
May 12-19 Observations and Causes of Seismic Activity, Suzdal, USSR. (I. M. Chersanov, Institute of Physics of the Earth, Bolshaya Gruzinskaya 10, Moscow 123810, USSR.)
May 12-22 Fifth International Symposium on Geophysics, Urbana, Ill. (Meetings, AGU, 2000 Florida Ave., N.W., Washington, DC 20009.)
May 17-22 Symposium on Remote Sensing and

Mineral Exploration, Ottawa, Ontario, Canada. Sponsors, Committee on Space Research (COSPAR) of the International Committee of Scientific Unions (ICSU), (W. D. Carter, EOS Office, U.S.G.S. (MS 730), Reston, VA 22092.)
May 17-June 3 24th Plenary Meeting of COSPAR, Ottawa, Ontario, Canada. (T. W. McGrath, Executive Member, Local Organizing Committee, XXIV COSPAR, Conference Secretariat, National Research Council, Ottawa, Ontario K1A 0R6, Canada.)
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